8.8 The Group 7 elements

1 Complete the word equations for the reactions of the Group 7 elements with metals.

a iron + ..chlorine..... → iron chloride
b iron + bromine → ....iron bromide..
c sodium + ..iodine..... → sodium iodide
d potassium + bromine → ..potassium bromide
e lithium + ...chlorine..... → lithium chloride
f sodium + chlorine → ..sodium chloride...

2 Make up four sentences using the phrases in the table. Each sentence must include a phrase from each column.

There are	weak forces strong forces	between the two atoms of a chlorine molecule. between a chlorine molecule and its neighbours. between the two atoms of an iodine molecule. between an iodine molecule and its neighbours.

There are weak forces between the two atoms of a chlorine molecule.
There are strong forces between a chlorine molecule and its neighbour
There are weak forces between the two atoms of an iodine molecule
There are strong forces between an iodine molecule and its neighbo

3 For the sentences A–H below:

- Write *1* next to the sentences that are true for Group 1.
- Write *7* next to the sentences that are true for Group 7.
- Write *B* next to the sentences that are true for both Groups 1 and 7.

A Going down this group, boiling point increases. ...7.....
B Going up this group, the reactions of the elements get more vigorous. ...1.....
C Going down this group, the number of protons in atoms of the elements increases. ...B.....
D Atoms of the elements in this group form molecules made up of two atoms. ...1.....
E Going up this group, the colours of the elements get lighter. ...7.....
F The elements at the top of this group are gases. .....
G The elements in this group conduct electricity. ...X.7....
H All the atoms of the elements in this group have the same number of electrons in the outermost shell. B.....

E

Write *M* next to the pair of elements that you predict will react most vigorously.
Write *L* next to the pair of elements that you predict will react least vigorously.

a sodium and iodine ...L.....
b lithium and iodine ...L.....
c potassium and fluorine ...M.....
d potassium and bromine ...M.....
e sodium and bromine ...L.....

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