

Chemistry 12 - Review of Significant Digits

TOTAL 28

The rules for zeros in significant digits are as follows:

- All zeros between non-zero digits are significant.
  - Zeros at the beginning of a number (eg. 0.0095) are NOT SIGNIFICANT!  
If the number 0.0095 was written in scientific notation, it would be:  $9.5 \times 10^{-3}$ . The exponent is not counted as significant so this number has 2 significant digits.
  - Zeros on the right side of a number (at the end) are significant if the DECIMAL POINT is shown.  
eg) 50.00 has 4 significant digits  
43.0 has 3 significant digits  
20. has 2 significant digits  
100. has 3 significant digits
  - Zeros to the left of an UNDERSTOOD decimal point are NOT significant.  
eg) 300 has 1 significant digit  
10 000 has 1 significant digit  
12 320 has 4 significant digits  
420 has 2 significant digits
1. Find the number of significant digits in each of the following measurements:

- |                                  |                                   |
|----------------------------------|-----------------------------------|
| a) 3.4005 ..... 5                | f) $9.080 \times 10^{-3}$ ..... 4 |
| b) 2580 ..... 3                  | g) 1.00 ..... 3                   |
| c) $3.20 \times 10^{-2}$ ..... 3 | h) 0.0027890 ..... 5              |
| d) 0.000308 ..... 3              | i) 320 000 ..... 2                |
| e) 23.000 ..... 5                | j) 9 ..... 1                      |

# Answer Key Instructional Fair Inc Balance Equation

**Michael Rajnik**



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