

SCIENCE 8 – DENSITY CALCULATIONS WORKSHEET

NAME: Key

- 1) A student measures the mass of an 8 cm^3 block of brown sugar to be 12.9 g . What is the density of the brown sugar?

$$m = 12.9 \text{ g}$$

$$V = 8 \text{ cm}^3$$

$$D = ?$$

$$D = \frac{m}{V}$$

$$D = \frac{12.9 \text{ g}}{8 \text{ cm}^3}$$

$$D = 1.6125 \text{ g/cm}^3$$

- 2) A chef fills a 50 mL container with 43.5 g of cooking oil. What is the density of the oil?

$$V = 50 \text{ mL}$$

$$m = 43.5 \text{ g}$$

$$D = \frac{m}{V}$$

$$D = \frac{43.5 \text{ g}}{50 \text{ mL}}$$

$$D = .87 \text{ g/mL}$$

- 3) Calculate the mass of a liquid with a density of 2.5 g/mL and a volume of 15 mL .

$$D = 2.5 \text{ g/mL}$$

$$V = 15 \text{ mL}$$

$$2.5 \text{ g/mL} = \frac{m}{15 \text{ mL}}$$

$$m = (2.5)(15)$$

$$m = 37.5 \text{ g}$$

- 4) Calculate the volume of a liquid with a density of 5.45 g/mL and a mass of 65 g .

$$D = 5.45 \text{ g/mL}$$

$$m = 65 \text{ g}$$

$$5.45 = \frac{65}{V}$$

$$V = \frac{65}{5.45}$$

$$V = 11.93 \text{ mL}$$

- 5) A machine shop worker records the mass of an aluminum cube as 176 g . If one side of the cube measures 4 cm , what is the density of the aluminum?

$$m = 176 \text{ g}$$

$$V = L \times W \times H$$

$$D = \frac{176 \text{ g}}{64 \text{ cm}^3}$$

$$D = 2.75 \text{ g/cm}^3$$

- 6) A scientist performing a demonstration finds that a piece of cork displaces 23.8 mL of water. The piece of cork has a mass of 5.8 g . What is the density of the cork?

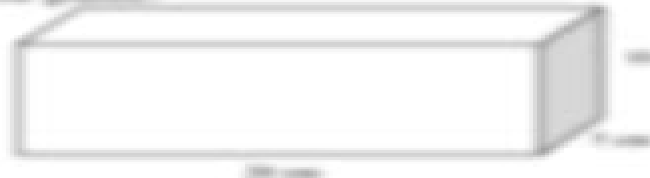
$$m = 5.8 \text{ g}$$

$$V = 23.8 \text{ mL}$$

$$D = \frac{5.8}{23.8}$$

$$D = .24 \text{ g/mL}$$

- 7) A carpenter begins work on the following block of granite that weighs 2700 g . What is the density of the granite?



$$m = 2700 \text{ g}$$

$$V = L \times W \times H$$

$$= 20 \times 10 \times 8$$

$$= 1600 \text{ cm}^3$$

$$D = \frac{2700}{1600}$$

$$D = 2.7 \text{ g/cm}^3$$

- 8) A piece of PVC plumbing pipe displaces 48 mL when placed into a container of water. If the pipe has a mass of 78 g , what is the density of PVC?

$$V = 48 \text{ mL}$$

$$m = 78 \text{ g}$$

$$D = \frac{78}{48}$$

$$D = 1.63 \text{ g/mL}$$

- 9) A solid magnesium flake has a mass of 1700 g and a volume of 743 cm^3 . What is the density of the magnesium?

$$m = 1700 \text{ g}$$

$$V = 743 \text{ cm}^3$$

$$D = \frac{1700}{743}$$

$$D = 2.29 \text{ g/cm}^3$$

Answer Key For Density Determination

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Standards**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

Answer Key For Density Determination:

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