

CH- ALDEHYDE, KETONES & CARBOXYLIC ACID

PRACTICE QUESTIONS

Q1. Convert Propene to Acetone.
Q2. Do the following conversions in not more than two steps: a) Ethyl Benzene to Benzoic Acid b) Propanone to Propene
Q3. Give reasons: a) Propanone is less reactive than ethanal towards nucleophilic addition reactions. b) $(\text{CH}_3)_2\text{CH-CHO}$ undergoes aldol condensation whereas $(\text{CH}_3)_3\text{C-CHO}$ does not.
Q4. Give a chemical test to distinguish between Butanal and Butan-2-one.
Q5. Convert Acetophenone to Benzoic Acid.
Q6. Give reasons: a) The α -hydrogen atoms of aldehydes and ketones are acidic in nature b) Propanone is less reactive than ethanal toward addition of HCN.
Q7. Convert: a) Toulene to Benzaldehyde b) Acetone to Propane
Q8. Give a chemical test to distinguish between Propanal and Propanone?
Q9. Arrange the following in increasing order of their reactivity towards nucleophilic addition reaction: a) $\text{C}_6\text{H}_5\text{COCH}_3$, $\text{CH}_3\text{-CHO}$, CH_3COCH_3 b) CH_3CHO , $\text{C}_6\text{H}_5\text{CHO}$, HCHO c) CH_3COCH_3 , $\text{C}_6\text{H}_5\text{COC}_6\text{H}_5$, CH_3CHO
Q10. Write the structure of the main products when propanal ($\text{CH}_3\text{-CH}_2\text{-CHO}$) reacts with the following reagents: a) Zn-Hg/conc. HCl b) $\text{H}_2\text{N-OH/H}^+$ c) HCN d) CH_3MgBr and then H_3O^+ e) $\text{H}_2\text{N-NHCONH}_2/\text{H}^+$ f) LiAlH_4
Q11. Write the products formed when ethanal reacts with the following reagents: a) CH_3MgBr and then H_3O^+ b) Zn-Hg/conc. HCl c) $\text{C}_6\text{H}_5\text{CHO}$ in the presence of dilute NaOH
Q12. Out of $\text{CH}_3\text{CH}_2\text{-CO-CH}_2\text{-CH}_3$ and $\text{CH}_3\text{CH}_2\text{-CH}_2\text{-CO-CH}_3$, which gives iodoform test?

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Career Point Kota



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Table of Contents Aldehyde Ketone And Carboxylic Acid Multiple Choice

1. Understanding the eBook Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - The Rise of Digital Reading Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Advantages of eBooks Over Traditional Books
2. Identifying Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - User-Friendly Interface
4. Exploring eBook Recommendations from Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Personalized Recommendations

- Aldehyde Ketone And Carboxylic Acid Multiple Choice User Reviews and Ratings
- Aldehyde Ketone And Carboxylic Acid Multiple Choice and Bestseller Lists
- 5. Accessing Aldehyde Ketone And Carboxylic Acid Multiple Choice Free and Paid eBooks
 - Aldehyde Ketone And Carboxylic Acid Multiple Choice Public Domain eBooks
 - Aldehyde Ketone And Carboxylic Acid Multiple Choice eBook Subscription Services
 - Aldehyde Ketone And Carboxylic Acid Multiple Choice Budget-Friendly Options
- 6. Navigating Aldehyde Ketone And Carboxylic Acid Multiple Choice eBook Formats
 - ePub, PDF, MOBI, and More
 - Aldehyde Ketone And Carboxylic Acid Multiple Choice Compatibility with Devices
 - Aldehyde Ketone And Carboxylic Acid Multiple Choice Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Highlighting and Note-Taking Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Interactive Elements Aldehyde Ketone And Carboxylic Acid Multiple Choice
- 8. Staying Engaged with Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Aldehyde Ketone And Carboxylic Acid Multiple Choice
- 9. Balancing eBooks and Physical Books Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Aldehyde Ketone And Carboxylic Acid Multiple Choice
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Setting Reading Goals Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aldehyde Ketone And Carboxylic Acid Multiple Choice
 - Fact-Checking eBook Content of Aldehyde Ketone And Carboxylic Acid Multiple Choice

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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