

Gases Assessment Review

Name _____

Skill - Student will accurately describe the relationship between pressure, volume, temperature and moles.

1. For each of the following scenarios, describe the change in the VOLUME of the gas. Label each scenario as exhibiting a direct or inverse relationship.

- Decrease in pressure on the gas
- Decrease in amount (mole) of gas
- Decrease in temperature of gas

Skill - Student will accurately calculate a 2 variable gas law problem.

2. A sample of carbon dioxide gas that has a volume of 5.2 L @ 25°C is heated to 32°C at a constant pressure. Calculate the new volume. **What equation must you use? Show your work.**

Skill - Student will accurately calculate a 3 variable gas law problem (not STP).

3. A sample of Kr gas with a volume of 4.0 L at a temp of -13°C and a pressure of 0.945 atm is heated to 56°C and a pressure of 1.18 atm. Calculate the final volume. **What equation must you use? Show your work.**

4. Hydrogen gas, starting @ 27 °C, at a pressure of 1.25 atm and a volume of 3.2 L is cooled to 2 °C, resulting in a volume of 2.5 L. What is the final pressure of H₂ in the system? **Show your work.**

Skill - Student will accurately calculate using the ideal gas law.

5. How many grams of chlorine gas are contained in 790.0 mL at 41.0 °C and 750.0 torr pressure? **What equation must you use? Show your work.**

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6. A 2.0L flask contains a mixture of nitrogen gas and oxygen gas at 35°C. The pressure from oxygen gas is 0.95 atm, and the mixture is known to contain 0.050mol nitrogen gas.

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