

States of Matter - PhET Simulator

Instructions: Open the States of Matter: Basics simulator via the PhET website or app.

Choose the "States" option. Change the temperature setting to "Celsius" above the thermometer.

- 1) Use the menu on the right side to select **Water** atoms, then choose the **Solid** state of matter. Draw a diagram of solid water below, and then describe the molecules in the next space.

Diagram	Description
	The water molecules are all very close together and vibrating slightly. Very little movement occurring.

- 2) Use the slider on the bottom to add heat (hold the fire upwards to heat). Note the thermometer changing as heat is added. What happens to the water molecules as heat is increased?

As heat is added, the water molecules begin to move more and separate from each other. They move farther and farther apart.

Experiment with the water molecules by adding and removing heat. Note the phase changes.

- 3) What is the approximate the melting point of water in Celsius?

(**Hint:** Check the temperature when this phase change occurs.)

0 degrees Celsius

- 4) How does the behavior of the water molecules below the melting point differ from water molecules that are above the melting point?

Below the melting point, molecules move slowly and close together. Above the melting point, molecules separate and move faster.

Answer Phet Lab States Of Matter

Mark Anderson, Olly Lewis



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Table of Contents Answer Phet Lab States Of Matter

1. Understanding the eBook Answer Phet Lab States Of Matter
 - The Rise of Digital Reading Answer Phet Lab States Of Matter
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Phet Lab States Of Matter
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Phet Lab States Of Matter
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Phet Lab States Of Matter
 - Personalized Recommendations
 - Answer Phet Lab States Of Matter User Reviews and Ratings
 - Answer Phet Lab States Of Matter and Bestseller Lists
5. Accessing Answer Phet Lab States Of Matter Free and Paid eBooks

- Answer Phet Lab States Of Matter Public Domain eBooks
- Answer Phet Lab States Of Matter eBook Subscription Services
- Answer Phet Lab States Of Matter Budget-Friendly Options
- 6. Navigating Answer Phet Lab States Of Matter eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Phet Lab States Of Matter Compatibility with Devices
 - Answer Phet Lab States Of Matter Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Phet Lab States Of Matter
 - Highlighting and Note-Taking Answer Phet Lab States Of Matter
 - Interactive Elements Answer Phet Lab States Of Matter
- 8. Staying Engaged with Answer Phet Lab States Of Matter
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Phet Lab States Of Matter
- 9. Balancing eBooks and Physical Books Answer Phet Lab States Of Matter
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Phet Lab States Of Matter
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Phet Lab States Of Matter
 - Setting Reading Goals Answer Phet Lab States Of Matter
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Phet Lab States Of Matter
 - Fact-Checking eBook Content of Answer Phet Lab States Of Matter
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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