

| Page 4 | Mark Scheme: Teachers' version<br>IGCSE – October/November 2009 | Syllabus<br>0620 | Paper<br>32 |
|--------|-----------------------------------------------------------------|------------------|-------------|
|--------|-----------------------------------------------------------------|------------------|-------------|

- (b) zinc is more reactive [1]  
 it loses electrons and forms ions in preference to iron [1]  
 zinc corrodes not iron [1]  
**NOT** zinc rusts

**OR** zinc loses electrons and forms ions [1]  
 the electrons move on to the iron [1]  
 the iron cannot be oxidised **or** it cannot rust **or** it cannot lose electrons [1]  
**CREDIT** correct Chemistry that includes the above ideas

- (c) (i) zinc atoms change into ions, (the zinc dissolves) [1]  
 copper(II) ions change into atoms, (becomes plated with copper) [1]

- (ii) ions [1]  
 electrons [1]

[Total: 10]

- 4 (a) diffusion [1]  
 different  $M_r$  **or** ozone molecules heavier than oxygen molecules  
**or** different densities **or** oxygen molecules move faster than ozone molecules [1]  
**NOT** oxygen is lighter **or** ozone heavier

**OR** fractional distillation [1]  
 they have different boiling points [1]

- (b) (i) from colourless (solution) [1]  
 to brown (solution) [1]

- (ii)  $I^-$  loses electrons (it is oxidised) [1]

- (iii) they are accepted by ozone [1]  
**or** ozone is an electron acceptor

- (c) (i) water  
 carbon dioxide  
 sulfur dioxide  
 all three [2]  
 any two [1]

- (ii) correct structural skeleton [1]  
**COND** 4bp around both carbon atoms [1]  
 2bp and 2bp around sulfur atom [1]

[Total: 11]

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**TD Snyder**



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