

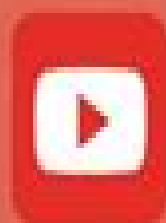
IGCSE A* MODEL ANSWERS

0620 CHEMISTRY

MAY/JUNE - 2024

PAPER - 61

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Table 1 shows the tests and the student's observations for solid P.

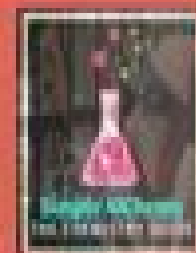
Tests	Observations
Test 1 Heat about half of solid P in boiling tube with test tube in fumes of HCl .	<u>Ammonia fumes</u> The white solid forms a colourless fume which diffuses out from the boiling tube and condensation is seen at the top of the boiling tube. After a while the colourless fume becomes a white solid.
Test 2 The remaining solid P is dissolved in water to form solution P. Solution P is divided into three portions. To the first portion of solution P, a few drops of sodium hydroxide solution are added and then it is warmed. Repeat the procedure and then using red litmus paper at the mouth of the boiling tube.	<u>Ammonia fumes</u> no H_2S gas no H_2SO_4 no change The fumes red litmus paper remains red.
Test 3 To the second portion of solution P, a few drops of dilute hydrochloric acid are added and then it is warmed.	<u>White precipitate</u> SO_4^{2-}
Test 4 To the third portion of solution P, with a few drops of acidified aqueous potassium manganate(VII).	The solution becomes light purple. SO_3^{2-}

30. The observations in Test 1 show that solid P is hydroxide.

31. Suggest a chemical test to show that the condensation at the top of the boiling tube contains ammonia.

anhydrous blue cobalt chloride paper
turns pink

anhydrous white crumbly solid



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Mike Jess

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