Page 3	Mark Scheme	Syllabus	Paper
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- (e) test (1)
  anhydrous copper sulfate or cobalt chloride(paper) (1)
  result (1)
  turns blue or pink (1)
  ignore: original colour

  [2]
- 4 (a) table of results for experiment 1

initial and final volumes and differences completed correctly (1)
15.7, 0.0 and 15.7
to 1 decimal place (1)
allow: 2 decimal places

- (b) table of results for experiment 2 initial and final volumes completed correctly (1) 47.3 and 15.9 differences completed correctly (1) 31.4 [2]
- (c) iron / Fe (1) (II) / 2+ (1) exidised / reacts with air / to iron(III) (1) [3]
- (d) (i) colourless to pink / purple (1) [1] not: clear allow: reverse
  - (ii) not an acid and alkali reaction / potassium manganate is coloured / owtte / indicator not needed / a colour change already occurs / potassium manganate acts as an indicator (1)
- (e) (i) experiment 2 (1) [1]
  - (ii) experiment 2 2× volume experiment 1 [1]
  - (iii) solution E more concentrated / stronger (1) or converse 2 × as concentrated (2) [2]
- (f) half value from table result for Experiment 2 / 15.7 cm<sup>3</sup> (1) half volume of E used (1) [2]

[2]

(g) advantage easy to use / quick / convenient (1) disadvantage not accurate / owtte (1)

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