

## Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O<sub>2</sub> 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



- 21 mol O<sub>2</sub> 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



- 45 mol H<sub>2</sub>O 3. How many moles of water are produced from 22.5 moles of oxygen?



- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



- 21 mol Cl<sub>2</sub> 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



# Answer Sheet To Chemistry Quiz Review Stoichiometry

**Research and Education Association**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the text 'Research and Education Association'.

## **Answer Sheet To Chemistry Quiz Review Stoichiometry:**

## **Answer Sheet To Chemistry Quiz Review Stoichiometry** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the power of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Answer Sheet To Chemistry Quiz Review Stoichiometry**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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