

5. On the graph paper provided, plot the volume-temperature values used in Calculation 4. Temperature data **must** be in °C. Draw a straight line between the two plotted points and extrapolate (extend) the line so that it crosses the temperature axis.

QUESTIONS AND PROBLEMS

1. (a) In the experiment, why are the water levels inside and outside the flask equalized before removing the flask from the cold water?

- (b) When the water level is higher inside than outside the flask, is the gas pressure in the flask higher than, lower than, or the same as, the atmospheric pressure? (specify which)

2. A 125-mL sample of dry air at 230°C is cooled to 100°C at constant pressure. What volume will the dry air occupy at 100°C?

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3. A 250 mL container of a gas is at 150°C. At what temperature will the gas occupy a volume of 125 mL, the pressure remaining constant?

10

4. (a) An open flask of air is cooled. Answer the following:

1. Under which conditions, before or after cooling, does the flask contain more gas molecules?

2. Is the pressure in the flask at the lower temperature the same as, greater than, or less than the pressure in the flask before it was cooled?

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