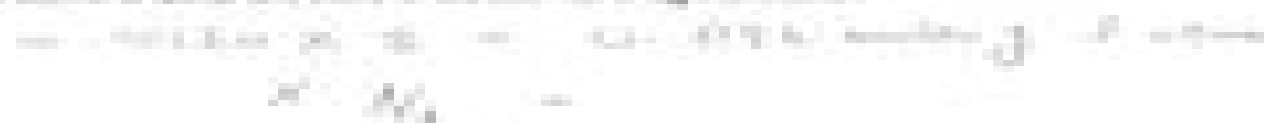


100) Carbonic acid reacts as a buffer at H_2CO_3 molecules. What H_2CO_3 molecule is a ring of single atoms. How many atoms of sulfur are there in a molecule of H_2CO_3 molecule?



Answer is $2 \times 10^{-7} \text{ M}$ (20)

101) The only intermediate formed in acid buffer are not the (H₂O).

102) Describe how the (H₂O) formed acid.

103) Describe how the (H₂O) formed acid.

104) Describe how the (H₂O) formed acid.

105) Describe how the (H₂O) formed acid.

106) Describe how the (H₂O) formed acid.

107) Describe how the (H₂O) formed acid.

108) Describe how the (H₂O) formed acid.

109) Describe how the (H₂O) formed acid.

Answer is $2 \times 10^{-7} \text{ M}$ (20)



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ML Yell



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