

Chemistry Study Guide

Periodic Table of the Elements

H 1 1.008																	He 2 4.003	
Li 3 6.941	Be 4 9.012											B 5 10.811	C 6 12.011	N 7 14.007	O 8 15.999	F 9 18.998	Ne 10 20.180	
Na 11 22.990	Mg 12 24.305											Al 13 26.982	Si 14 28.086	P 15 30.974	S 16 32.06	Cl 17 35.45	Ar 18 39.948	
K 19 39.098	Ca 20 40.078	Sc 21 44.956	Ti 22 47.88	V 23 50.942	Cr 24 51.996	Mn 25 54.938	Fe 26 55.845	Co 27 58.933	Ni 28 58.69	Cu 29 63.546	Zn 30 65.38	Ga 31 69.723	Ge 32 72.63	As 33 74.922	Se 34 78.96	Br 35 79.904	Kr 36 83.80	
Rb 37 85.468	Sr 38 87.62	Y 39 88.906	Zr 40 91.224	Nb 41 92.906	Mo 42 95.94	Tc 43 98.906	Ru 44 101.07	Rh 45 102.91	Pd 46 106.42	Ag 47 107.87	Cd 48 112.41	In 49 114.82	Sn 50 118.71	Sb 51 121.76	Te 52 127.6	I 53 126.91	Xe 54 131.29	
Cs 55 132.91	Ba 56 137.33			Hf 72 178.49	Ta 73 180.95	W 74 183.84	Re 75 186.21	Os 76 190.23	Ir 77 192.22	Pt 78 195.08	Au 79 196.97	Hg 80 200.59	Tl 81 204.38	Pb 82 207.2	Bi 83 208.98	Po 84 209	At 85 210	Rn 86 222
Fr 87 223	Ra 88 226			Rf 104 261	Db 105 262	Sg 106 266	Bh 107 264	Hs 108 277	Mt 109 268	Ds 110 271	Rg 111 272	Cn 112 285	Uut 113 284	Fl 114 289	Uup 115 288	Lv 116 293	Uus 117 286	Uuo 118 294
Lanthanide series La 57 138.91 Ce 58 140.12 Pr 59 140.91 Nd 60 144.24 Pm 61 144.91 Sm 62 150.36 Eu 63 151.96 Gd 64 157.25 Tb 65 158.93 Dy 66 162.50 Ho 67 164.93 Er 68 167.26 Tm 69 168.93 Yb 70 173.05 Lu 71 174.97																		
Actinide series Ac 89 227 Th 90 232.04 Pa 91 231.04 U 92 238.03 Np 93 237 Pu 94 244 Am 95 243 Cm 96 247 Bk 97 247 Cf 98 251 Es 99 252 Fm 100 257 Md 101 258 No 102 259 Lr 103 262																		

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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