

Peugeot

405

WORKSHOP MANUAL



1987-1997

Peugeot 405 Workshop Service Repair Manual

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 $\text{C}_2\text{H}_2(\text{g}) + 2 \text{ H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) \quad \Delta h = -84.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = -242 \text{ kJ}$

co₂ g 3 h 2 o g Δh 283 kj g 1 c₂h₆ g

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web thus for reaction 9 2 in which 1 mole of H_3O^+ and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta h = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta h_{\text{products}}$

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like to use pouring medium and water to do this the pouring medium helps with thinning out the paint and prevents it from cracking while it dries i like to mix extra water in with the medium because it will give the

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