

Academic Chemistry

Mr. Gensits

Section Review 7.2 Answer Key

1. electrostatic forces
2. oppositely
3. ionic bonds
4. neutral
5. formula unit
6. crystals
7. high
8. large
9. stable
10. molten

11. AT
12. ST
13. AT
14. ST
15. NT

16. b
17. e
18. c
19. d
20. a

21. Ionic bonds are the electrostatic forces of attraction that bind oppositely charged ions together. In an ionic compound, the positive charges of the cations equal the negative charges of the anions.

22. When ionic compounds are melted, the orderly crystal structure breaks down. Each ion is then free to move throughout the molten mass. If a voltage is applied, cations will migrate to one electrode and anions will migrate to the other. This movement of ions means that there is a flow of electricity between the two electrodes. When ionic compounds dissolve in water, the ions are free to move. Thus, aqueous solutions of ionic compounds also conduct electricity.

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Daniela Niemeyer



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