

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
  - Two nonmetal atoms bonded together
  - Two metal atoms bonded together
  - A metal and a nonmetal atom bonded together
  - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
16. (4 pts) How many total atoms are in this molecule:  $C_6H_{12}OH$ ?
- 3
  - 4
  - 19
  - 20
  - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is  $C_5H_{10}Cl_x$ , what must  $x$  be?
- 1
  - 2
  - 3
  - 4
  - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water?  $C_3H_6ClF$   $C_6H_5CH_3$   $CS_2$   $BaSO_4$
- $C_3H_6ClF$
  - $C_6H_5CH_3$
  - $CS_2$
  - $BaSO_4$
  - $CS_2$  and  $BaSO_4$

# Answers For Chemistry Test Module 13

**Jessica J Manson**



## Answers For Chemistry Test Module 13:

**Nelson Modular Science** Paul Collinson, Paul Collison, 2005-03 The Nelson Modular Science series is made up of three books divided into single double and triple award modules presented in an accessible format Book 1 covers the six single award and one coursework modules Book 2 contains six double award modules and Book 3 covers the six triple award modules Each module is covered in self contained units This teacher s file includes practical support sheets and addresses Sc1 investigations Works sheets are provided to integrate the use of ICT throughout science Additional GCSE style questions and modular tests should enhance learning and recall of information      **Student Study Guide to Accompany Statistics Alive! 2e by Wendy J. Steinberg** Wendy J. Steinberg, Matthew Price, 2010-11-11 This affordable student study guide and workbook to accompany Wendy Steinberg s Statistics Alive text will help students get the added review and practice they need to improve their skills and master their Introduction to Statistics course      Revise for Edexcel Modular Science Graham Booth, 2002 An exact match to the structure and content of the new Edexcel Modular specification Prepare your students for both the end of module tests and the terminal exams Provide lots of exam style questions to make sure students get plenty of practice Offer hints and tips to help students prepare for their terminal exams      El-Hi Textbooks in Print , 1981      *Problems and Problem Solving in Chemistry Education* Georgios Tsaparlis, 2021 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner      *Access to Chemistry* Avril Higton, Mike Clemmet, Elaine Golding, Alan V Jones, 2007-10-31 This text has been specifically designed to prepare people with previously limited chemical knowledge for entrance into science related courses such as Foundation and Access courses which involve chemistry in higher education Until now there have been no texts available for use on these courses and this book fills that gap Access to Chemistry effectively forms a self study course which is split into separate modules and units covering the full spread of

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