Kinetics Worksheet 1

3.	For the reaction:	A + 2B	\rightarrow 3 C $+$ D	Ġ
	the following data were obtained,			
	271 6101	EA.3	COL	

THEORY AND S	AS KNOWN THEORY		
mun	[A]	[8]	rate M·s·1
1	0.16	0.36	2.67
2	0.16	0.09	0.67
3	0.32	0.24	7.11
4	0.08	0.09	0.168
5	0.24	0.12	2.0
6	0.02	1.35	7

$$\frac{(0.16)^{1}}{(0.08)^{1}} = \frac{0.67ms^{-1}}{0.168ms^{-1}}$$

$$(2)^{1} = 4$$

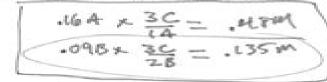
$$y + 1 = 2 = 204$$

$$y = 2 = 204$$

$$y = 2 = 204$$

(C) Calculate the rate of the reaction when [A] = 0.02 M and [B] = 1.35 M.

(D) Assume that the reaction goes to completion. Under the conditions specified in the second run, what would the final [C] be?





Answers Kinetics Worksheet 1

Arshad Iqbal



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