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Quiz & Worksheet - Isotopes and Average Atomic Mass

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1. How are isotopes of the same element similar?

d. They have the same particles in the nucleus.

c. They are equally abundant in nature.

C. They have the same mass.

c. They have the same number of neutrons.

They have the same number of protons.

2. How are isotopes of the same element different?

They will have undergo different chemical reactions.

C They have different numbers of protons.

c. They have different numbers of neutrons.

c. They will have different locations on the periodic table.

a They will have different numbers of electrons.

3. How many neutrons does the most common isotope of hydrogen have?

0 3

@ 2

0

0 0

65

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Answers Isotopes And Atomic Mass Phet Sheet

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Answers Isotopes And Atomic Mass Phet Sheet:

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