



Chemistry

Vocab Puzzle



The Periodic Table Vocabulary

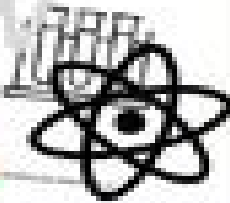
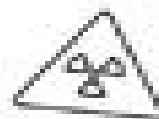
1. **Alkaline Earth Metals** - elements in the second column of the periodic table.
2. **Alkali Metals** - the elements of group 1 of the periodic table with low melting points.
3. **Halogens** - the elements of group 17 of the periodic table.
4. **Electronegativity** - the tendency for an atom of an element to attract electrons when forming a chemical bond, affected by the distance electrons travel from the nucleus.
5. **Group** - vertical column in the periodic table consisting of elements with similar properties.
6. **Period** - horizontal row in the periodic table consisting of elements with similar properties.
7. **Transition Metals** - elements in the middle of the periodic table.
8. **Non-metals** - elements on the right side of the periodic table.
9. **Metals** - elements on the left side of the periodic table.
10. **Lanthanide Series** - a series of elements below the main body of the periodic table.
11. **Actinide Series** - a series of elements below the lanthanide series.
12. **Isotopes** - atoms of an element with the same number of protons but different numbers of neutrons.
13. **Atomic Number** - the number of protons in an atom.
14. **Mass Number** - the sum of protons and neutrons in an atom.
15. **Chemical Bond** - the force that holds atoms together in a molecule or compound.
16. **Chemical Reaction** - a process where one or more substances are converted into one or more different substances.
17. **Chemical Equation** - a symbolic representation of a chemical reaction.
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26. **Chemical Mixture** - a combination of two or more different substances that are not chemically bonded.
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50. **Chemical Real Gas Equation** - an equation that takes into account the volume of gas molecules and the forces between them.
51. **Chemical Van der Waals Equation** - an equation that describes the behavior of real gases.
52. **Chemical Maxwell-Boltzmann Distribution** - a distribution that describes the speeds of particles in a gas.
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Name: _____

1. **Alkaline Earth Metals** - elements with partially filled s orbitals, get oxidation of +2.
2. **Alkali Metals** - elements with partially filled s orbitals, get oxidation of +1.
3. **Halogens** - elements with partially filled p orbitals, get oxidation of -1.
4. **Electronegativity** - the tendency for an atom of an element to attract electrons when forming a chemical bond, affected by the distance electrons travel from the nucleus.
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The Periodic Table Vocabulary Puzzle

With list of definitions and word bank option!



Answers For Periodic Table Vocabulary Modern Chemistry Puzzle

M Carnoy



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