Stoichiometry: Mass - Mole Worksheet

Based on the following chemical equation, please answer the following questions!

Fe (SO₄), + 6 NaCl --> 2 FeCl, + 3 Na₂SO₄

Calculate the amount of moles of NaCl needed to produce 75 g of FeCl,.

9 FeCl,	1 mole FeCl,	mole NaCl
	9 FeCl,	mole FeCl

Calculate the amount of Fe, (SO,), in grams produce 215 g Na, SO,.

9 Na ₃ SO ₄	1 mole Na SO	mole Fe ₂ (SO ₄),	9 Fe ₂ (SO ₄),
	9 Na ₂ SO ₄	mole Na,SO,	1 mole Fe ₂ (SO ₄),

Calculate the amount in grams of FeCl, produced with 95g of Fe (SO₄),

9 Fe ₂ (SO ₄),	mole Fe ₂ (SO ₄),	mole FeCl,	9 FeCl,
	9 Fe,(SO,),	mole Fe(SO ₄),	1 mole FeCl,

Calculate the amount of moles of Fe₂(SO₄) needed to react with 123.5 g of NaCl

9 NaCl	1 mole NaCl	mole Fe(SO ₄),	
	9 NaCl	mole NaCl	

Based on the following chemical equation, please answer the following questions!

9 Ca(HCO _y) ₂	1 mole Ca(HCO ₂),	mole CaCO,
	9 Ca(HCO ₃),	mole Ca(HCO,),

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Arihant Experts

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Study Guide/Selected Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd Chemistry, Student Study Guide James E. Brady, Fred numbered end of chapter problems in the text book Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Denniston Katherine, 2006 Study Guide and Full Solutions Manual Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every

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