

16. Given the chemical formula of an ionic compound, list at least three necessary steps to give the correct name of that compound. (If needed, use a chemical formula of a compound from the table above as an example in listing the naming steps.)

CHEMICAL BONDING POGIL

Activity 4 – Predicting the correct chemical formula for ionic compounds formed from simple anions

Objective:

- To learn how to predict the correct number of cations and anions in a simple salt.

The Model:

Table 3

Cations	Anions	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Na^+	S^{2-}	Na_2S	sodium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Exploring the Model

1. What is the charge on the zinc ion?
2. What is the charge on the nitride ion?
3. What is the charge on the chloride ion?
4. What is the charge on the ionic compound, sodium chloride?
5. What is the charge on the ionic compound, sodium sulfide?
6. How many potassium ions are present in K_3N ?
7. What does the “2” stand for in the formula for ZnCl_2 ?

Exploring the Model

8. Sodium chloride is NaCl , and zinc chloride is ZnCl_2 . Why are there more chloride ions in the zinc compound?
9. Sodium chloride is NaCl , and sodium sulfide is Na_2S . Why are there more sodium ions in the sulfide compound?

Answers Chemical Bonding Pogil

Phillip Manning

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes **Organic Chemistry** Suzanne M. Ruder, The POGIL Project, 2015-12-29 **ORGANIC CHEMISTRY Foundations of Inorganic Chemistry** Gary Wulfsberg, 2017-11-02 Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence By covering virtually every topic in the test from the 2016 ACS Exams Institute this book will prepare your students for success The new book combines careful pedagogy clear writing beautifully rendered two color art and solved examples with a broad array of original chapter ending exercises It assumes a background in General Chemistry but reviews key concepts and also assumes enrollment in a Foundations of Organic Chemistry course Symmetry and molecular orbital theory are introduced after the student has developed an understanding of fundamental trends in chemical properties and reactions across the periodic table which allows MO theory to be more broadly applied in subsequent chapters Use of this text is expected to increase student enrollment and build students appreciation of the central role of inorganic chemistry in any allied field Key Features Over 900 end of chapter exercises half answered in the back of the book Over 180 worked examples Optional experiments demos Clearly cited connections to other areas in chemistry and chemical sciences Chapter opening biographical vignettes of noted scientists in Inorganic Chemistry Optional General Chemistry review sections Originally rendered two color illustrations throughout *Analytical Chemistry* Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more *Transforming Urban Education* Kenneth Tobin, Ashraf Shady, 2014-04-03 Transformations in Urban Education Urban Teachers and Students Working Collaboratively addresses pressing problems in urban education contextualized in research in New York City and nearby school districts on the Northeast Coast of the United States The schools and institutions involved in empirical studies range from elementary through college and include public and private schools alternative schools for dropouts and museums Difference is regarded as a resource for learning and equity issues are examined in terms of race ethnicity language proficiency designation as special education and gender

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Advances in Teaching Physical Chemistry Mark David Ellison,2008 This book brings together the latest perspectives and ideas on teaching modern physical chemistry It includes perspectives from experienced and well known physical chemists a thorough review of the education literature pertaining to physical chemistry a thorough review of advances in undergraduate laboratory experiments from the past decade in depth descriptions of using computers to aid student learning and innovative ideas for teaching the fundamentals of physical chemistry This book will provide valuable insight and information to all teachers of physical chemistry

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