Electron Configurations Worksheet - Answers

Write the complete ground state electron configurations for the following:

| 1) | lithium | 1s ² 2s ¹ |
|----|----------|---|
| 2) | oxygen | 1s ² 2s ² 2p ⁴ |
| 3) | calcium | 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² |
| 4) | titanium | 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ² |
| 5) | rubidium | 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ¹ |
| 6) | lead | 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ² 4d ¹⁰ 5p ⁶ 6s ² 4f ¹⁴ 5d ¹⁰ 6p ² |
| 7) | erbium | 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ² 4d ¹⁰ 5p ⁶ 6s ² 4f ¹² |

Write the abbreviated ground state electron configurations for the following:

| 8) | helium | 1s ² (this one cannot be abbreviated) | |
|-----|----------|--|--|
| 9) | nitrogen | [He] 2s ² 2p ³ | |
| 10) | chlorine | [Ne] 3s ² 3p ⁵ | |
| 11) | iron | [Ar] 4s ² 3d ⁶ | |
| 12) | zinc | [Ar] 4s ² 3d ¹⁰ | |
| 13) | barium | [Xe] 6s ² | |
| 14) | polonium | [Xe] 6s ² 4f ¹⁴ 5d ¹⁰ 6p ⁴ | |
| | | | |

Answers For Electron Configuration Pogil Worksheet

Terry C. Jones

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