

Chemistry - Review Chapters (415) 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur
6

b. calcium
2

c. chlorine
7

d. arsenic
5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10
Ne, NEON

b. 18
Ar, ARGON

c. 36
Kr, KRYPTON

d. 90
Th, THORIUM

7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl

b. Na and Li

c. Na and Cu

d. Na and Ne

9. Which of the following pairs of elements belong to the same group?

a. H and He

b. Li and Be

c. C and Pb

d. Ga and Ge

10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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C Cleary



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Table of Contents Answers For Prentice Hall Chemistry Review Module

1. Understanding the eBook Answers For Prentice Hall Chemistry Review Module
 - The Rise of Digital Reading Answers For Prentice Hall Chemistry Review Module
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Prentice Hall Chemistry Review Module
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Prentice Hall Chemistry Review Module
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Prentice Hall Chemistry Review Module
 - Personalized Recommendations
 - Answers For Prentice Hall Chemistry Review Module User Reviews and Ratings
 - Answers For Prentice Hall Chemistry Review Module and Bestseller Lists

5. Accessing Answers For Prentice Hall Chemistry Review Module Free and Paid eBooks
 - Answers For Prentice Hall Chemistry Review Module Public Domain eBooks
 - Answers For Prentice Hall Chemistry Review Module eBook Subscription Services
 - Answers For Prentice Hall Chemistry Review Module Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Answers For Prentice Hall Chemistry Review Module Compatibility with Devices
 - Answers For Prentice Hall Chemistry Review Module Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Answers For Prentice Hall Chemistry Review Module
 - Highlighting and Note-Taking Answers For Prentice Hall Chemistry Review Module
 - Interactive Elements Answers For Prentice Hall Chemistry Review Module
8. Staying Engaged with Answers For Prentice Hall Chemistry Review Module
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Prentice Hall Chemistry Review Module
9. Balancing eBooks and Physical Books Answers For Prentice Hall Chemistry Review Module
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Prentice Hall Chemistry Review Module
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Prentice Hall Chemistry Review Module
 - Setting Reading Goals Answers For Prentice Hall Chemistry Review Module
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Prentice Hall Chemistry Review Module
 - Fact-Checking eBook Content of Answers For Prentice Hall Chemistry Review Module
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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