

Acids and Bases

Question 1: Choose the correct answer for each of the following questions:

- 1.1 The element present in all acids...
- A oxygen B carbon dioxide
C nitrogen D hydrogen
- 1.2 A sample of an acid in a beaker has a pH of 4. An alkali is added to it. What happens to the pH of the liquid in the beaker?
- A decreases B stays the same
C increases D neutral
- 1.3 Choose the incorrect combination:
- A HNO_3 – Nitric acid. B H_2SO_4 – sulphuric acid.
C HCl – Hydrochloric acid. D H_2CO_3 – carbonic acid.
- 1.4 The colour of blue litmus paper in an acid is ...
- A yellow B green
C blue D red
- 1.5 Which substance is always produced in the reaction between hydrochloric acid and lithium?
- A water B hydrogen gas
C oxygen gas D carbon dioxide gas
- 1.6 Which substance is always produced in the reaction between hydrochloric acid and sodium hydroxide?
- A water B hydrogen gas
C oxygen gas D precipitate
- 1.7 Which formula represents an acid?
- A KOH B KCl
C CH_3OH D CH_3COOH
- 1.8 The method of adding an acid to a base drop by drop:
- A neutralisation B precipitation
C evaporation D decomposition

Quest. Write the general reaction for an acid with the following compounds:

- 2.1 Base: _____
- 2.2 Carbonate: _____
- 2.3 Metal: _____

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