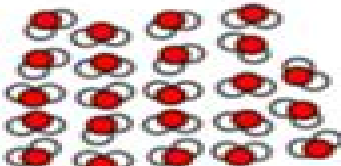


## **States of Matter – PhET Simulator**

**Instructions:** Open the *States of Matter: Basics* simulator via the PhET website or app. Choose the “**States**” option. Change the temperature setting to “**Celsius**” above the thermometer.

- 1) Use the menu on the right side to select **Water** atoms, then choose the **Solid** state of matter. Draw a diagram of **solid water** below, and then **describe the molecules** in the next space.

| Diagram                                                                           | Description                                                                                                           |
|-----------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------------|
|  | <p><b>The water molecules are all very close together and vibrating slightly. Very little movement occurring.</b></p> |

- 2) Use the slider on the bottom to **add heat** (hold the fire upwards to heat). Note the thermometer changing as heat is added. What happens to the **water molecules** as heat is increased?

**As heat is added, the water molecules begin to move more and separate from each other. They move further and further apart.**

Experiment with the water molecules by **adding and removing heat**. Note the phase changes.

- 3) What is the approximate the **melting point** of water in Celsius?

(**Hint:** Check the temperature when this phase change occurs.)

**0 degrees Celsius**

- 4) How does the behavior of the water molecules **below the melting point** differ from water molecules that are **above the melting point**?

**Below the melting point, molecules move slowly and close together. Above the melting point, molecules separate and move faster.**

# Answerkey Phet Simulation

**Victor M. Corman**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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