

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

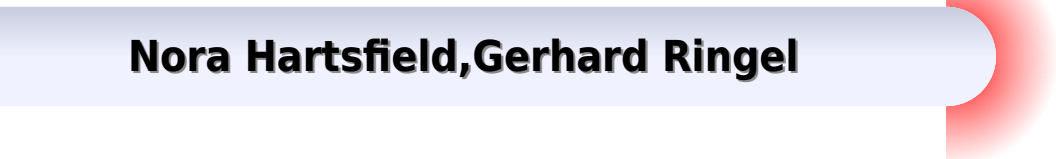
The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Answers For Modern Chemistry 9 2

Nora Hartsfield, Gerhard Ringel



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Modern Chemistry Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Pearls in Graph Theory** Nora Hartsfield,Gerhard Ringel,2013-04-15 Stimulating and accessible this undergraduate level text covers basic graph theory colorings of graphs circuits and cycles labeling graphs drawings of graphs measurements of closeness to planarity graphs on surfaces and applications and algorithms 1994 edition *Gauge Theory and Variational Principles* David Bleecker,2005-12-10 This text provides a framework for describing and organizing the basic forces of nature and the interactions of subatomic particles A detailed and self contained mathematical account of gauge theory it is geared toward beginning graduate students and advanced undergraduates in mathematics and physics This well organized treatment supplements its rigor with intuitive ideas Starting with an examination of principal fiber bundles and connections the text explores curvature particle fields Lagrangians and gauge invariance Lagrange s equation for particle fields and the inhomogeneous field equation Additional topics include free Dirac electron fields interactions calculus on frame bundle and unification of gauge fields and gravitation The text concludes with references a selected bibliography an index of notation and a general index **Algebraic Extensions of Fields** Paul J. McCarthy,2014-01-07 Graduate level coverage of Galois theory especially development of infinite Galois theory theory of valuations prolongation of rank one valuations more Over 200 exercises Bibliography clear unsophisticated and direct Math **Elementary Induction on Abstract Structures** Yiannis N. Moschovakis,2014-06-10 Well written research monograph recommended for students and professionals interested in model theory and definability theory Easy to use and a pleasure to read Bulletin of the American Mathematical Society 1974 edition *A User's Guide to Ellipsometry* Harland G. Tompkins,2013-03-21 This text on optics for graduate students explains how to determine material properties and parameters for inaccessible substrates and unknown films as well as how to measure extremely thin films Its 14 case studies illustrate concepts and reinforce applications of ellipsometry particularly in relation to the semiconductor industry and to studies involving corrosion and oxide growth A User s Guide to Ellipsometry will enable readers to move beyond limited turn key applications of ellipsometers In addition to its comprehensive discussions of the measurement of film thickness and optical constants in film it also considers the trajectories of the ellipsometric parameters Del and Psi and how changes in materials affect parameters This volume also addresses the use of polysilicon a material commonly employed in the microelectronics industry and the effects of substrate roughness Three appendices provide helpful references Theory of Flight Richard von Mises,2012-04-27 Mises classic avoids the formidable mathematical structure of fluid dynamics while conveying by often unorthodox methods a full

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