

Chemistry of Life

DOODLE NOTES

chemistry of life
BASIC CHEMISTRY REVIEW

atoms and life
All living things are made of atoms.
An atom is the smallest part of an element. Many different types of atoms exist and they are arranged in the Periodic table of elements. Atoms are made up of subatomic particles.

Label the structure of this oxygen atom:

Electron shell
Electron
Nucleon
Proton
Neutron

• State the charge and mass of each subatomic particle.

	Proton	Neutron	Electron
Charge	+	0	-
Mass	1	1	1/1836

• How many electrons are in the first shell? 2
• How many electrons are in the outer shell? 6
• What do we call electrons in the outer shell? Valence electrons

elements
An element is a substance that is made of one type of atom only. They cannot be broken down into simpler substances. There are 118 elements. There are 92 in the periodic table.

compounds
A compound is made of two or more types of atoms that are chemically bonded. They can be broken down chemically. There are 100 compounds. There are 100 in the periodic table.

mixtures
A mixture is a mix of two or more types of atoms that are not chemically bonded. They can be separated physically. There are 100 mixtures. There are 100 in the periodic table.

Hydrogen
Hydrogen is the simplest element. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

Helium
Helium is a noble gas. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

Lithium
Lithium is a metal. It is made of one type of atom only. It is a solid at room temperature. It is used in many chemical reactions.

Sodium
Sodium is a metal. It is made of one type of atom only. It is a solid at room temperature. It is used in many chemical reactions.

Chlorine
Chlorine is a non-metal. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

Oxygen
Oxygen is a non-metal. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

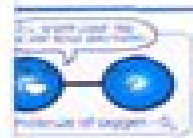
Carbon
Carbon is a non-metal. It is made of one type of atom only. It is a solid at room temperature. It is used in many chemical reactions.

Hydrogen chloride
Hydrogen chloride is a compound. It is made of two types of atoms. It is a gas at room temperature. It is used in many chemical reactions.

Sodium chloride
Sodium chloride is a compound. It is made of two types of atoms. It is a solid at room temperature. It is used in many chemical reactions.

Oxide
Oxide is a compound. It is made of two types of atoms. It is a solid at room temperature. It is used in many chemical reactions.

Water
Water is a compound. It is made of two types of atoms. It is a liquid at room temperature. It is used in many chemical reactions.



groups
Groups are vertical columns in the periodic table. They are numbered 1 to 18. They are used to predict the properties of elements.

Periods
Periods are horizontal rows in the periodic table. They are numbered 1 to 7. They are used to predict the properties of elements.

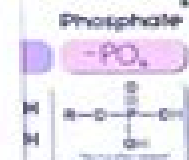
Metals
Metals are elements that are shiny, malleable, and good conductors of electricity. They are found on the left side of the periodic table.

Non-metals
Non-metals are elements that are not shiny, not malleable, and not good conductors of electricity. They are found on the right side of the periodic table.

Gases
Gases are elements that are in the gaseous state at room temperature. They are found on the right side of the periodic table.

Liquids
Liquids are elements that are in the liquid state at room temperature. They are found on the right side of the periodic table.

Solids
Solids are elements that are in the solid state at room temperature. They are found on the left side of the periodic table.



Acids
Acids are substances that release hydrogen ions (H^+) in aqueous solution. They are used in many chemical reactions.

Alkalies
Alkalies are substances that release hydroxide ions (OH^-) in aqueous solution. They are used in many chemical reactions.

Salts
Salts are ionic compounds that are formed from the reaction of an acid and a base. They are used in many chemical reactions.

Hydroxides
Hydroxides are compounds that contain the hydroxide ion (OH^-). They are used in many chemical reactions.

Oxides
Oxides are compounds that contain the oxide ion (O^{2-}). They are used in many chemical reactions.

Carbonates
Carbonates are compounds that contain the carbonate ion (CO_3^{2-}). They are used in many chemical reactions.

Sulfates
Sulfates are compounds that contain the sulfate ion (SO_4^{2-}). They are used in many chemical reactions.

Nitrates
Nitrates are compounds that contain the nitrate ion (NO_3^-). They are used in many chemical reactions.

Ammonium
Ammonium is a group of atoms that are bonded together. It is represented by the symbol NH_4^+ . It is used in many chemical reactions.



MULTIPLE VERSIONS & EXAMPLE

Answers Review Basic Units Of Life

Brigitte Niedzwiecki



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