

Gizmos: Types of Reactions

Activity B: Classifying reactions

Get the Gizmo ready:

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Introduction: Chemical equations show how **compounds** and **elements** react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or oxygen gas (O₂). A compound is a substance made of more than one kind of atom, such as water (H₂O) or table salt (NaCl).

Question: How are chemical reactions classified?

1. **Match:** Most chemical reactions can be classified as one of four types. Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.

____ One reactant is broken down into two or more products.

A. **Addition**

____ Two or more reactants combine to form one product.

B. **Decomposition**

____ Two compounds react to form two different compounds.

C. **Single replacement**

____ A compound reacts with an element to form a new compound and a different element.

D. **Double replacement**

2. **Practice:** Balance each of the chemical equations below. (Some equations may already be in balance.) In the space to the right, classify the reaction as an *addition*, *decomposition*, *single replacement*, or *double replacement* reaction.



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JE Gale

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Table of Contents Answer Sheets For Chemistry Gizmos

1. Understanding the eBook Answer Sheets For Chemistry Gizmos
 - The Rise of Digital Reading Answer Sheets For Chemistry Gizmos
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Sheets For Chemistry Gizmos
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Sheets For Chemistry Gizmos
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Sheets For Chemistry Gizmos
 - Personalized Recommendations
 - Answer Sheets For Chemistry Gizmos User Reviews and Ratings
 - Answer Sheets For Chemistry Gizmos and Bestseller Lists

5. Accessing Answer Sheets For Chemistry Gizmos Free and Paid eBooks
 - Answer Sheets For Chemistry Gizmos Public Domain eBooks
 - Answer Sheets For Chemistry Gizmos eBook Subscription Services
 - Answer Sheets For Chemistry Gizmos Budget-Friendly Options
6. Navigating Answer Sheets For Chemistry Gizmos eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Sheets For Chemistry Gizmos Compatibility with Devices
 - Answer Sheets For Chemistry Gizmos Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Sheets For Chemistry Gizmos
 - Highlighting and Note-Taking Answer Sheets For Chemistry Gizmos
 - Interactive Elements Answer Sheets For Chemistry Gizmos
8. Staying Engaged with Answer Sheets For Chemistry Gizmos
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Sheets For Chemistry Gizmos
9. Balancing eBooks and Physical Books Answer Sheets For Chemistry Gizmos
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Sheets For Chemistry Gizmos
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Sheets For Chemistry Gizmos
 - Setting Reading Goals Answer Sheets For Chemistry Gizmos
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Sheets For Chemistry Gizmos
 - Fact-Checking eBook Content of Answer Sheets For Chemistry Gizmos
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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