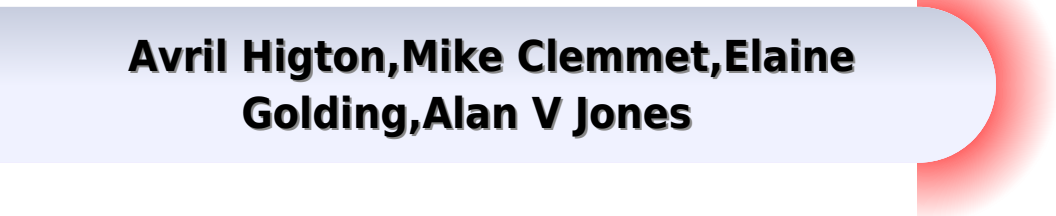


13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
 - Two nonmetal atoms bonded together
 - Two metal atoms bonded together
 - A metal and a nonmetal atom bonded together
 - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
16. (4 pts) How many total atoms are in this molecule: $C_6H_{12}OH$?
- 3
 - 4
 - 19
 - 20
 - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $C_5H_{10}Cl_x$, what must x be?
- 1
 - 2
 - 3
 - 4
 - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C_3H_6ClF $C_6H_5CH_3$ CS_2 $BaSO_4$
- C_3H_6ClF
 - $C_6H_5CH_3$
 - CS_2
 - $BaSO_4$
 - CS_2 and $BaSO_4$

Answers For Chemistry Test Module 13

**Avril Higton, Mike Clemmet, Elaine
Golding, Alan V Jones**



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1. This book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Answers For Chemistry Test Module 13
 - Chapter 3: Answers For Chemistry Test Module 13 in Everyday Life
 - Chapter 4: Answers For Chemistry Test Module 13 in Specific Contexts
 - Chapter 5: Conclusion
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5. In chapter 4, the author will scrutinize the relevance of Answers For Chemistry Test Module 13 in specific contexts. The fourth chapter will explore how Answers For Chemistry Test Module 13 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Answers For Chemistry Test Module 13. The final chapter will summarize the key points that have been discussed throughout the book.

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