



CHEMISTRY

Section Review

Objectives

1. Identify the different levels of study in chemistry.
2. Know your chemistry vocabulary words.
3. Identify experimental safety hazards.

Vocabulary

- | | | |
|-----------------------|------------------------|----------------------|
| a. matter | b. substance | c. pure chemistry |
| d. chemistry | e. chemical structure | f. applied chemistry |
| g. organic chemistry | h. inorganic chemistry | i. biochemistry |
| j. physical chemistry | | |

Part A Completion

Read the following sentences. Write the appropriate vocabulary word in the space provided. Write the chemical formula of the compound in the space provided after each compound name in parentheses.

Matter is anything that has _____ and _____.

Chemistry is the study of the _____ of matter and the

_____ and _____ changes it undergoes in nature and in the

laboratory. _____ chemistry studies changes that occur only

in chemical reactions. _____ study changes that occur in

physically changing or physically combining matter. _____

Chemistry is the study of the properties and relationships

_____ and _____, as well as the composition of matter.

_____ chemists determine the _____ and _____ that

control molecular structure and properties of chemical systems in the

laboratory. _____ study the structure of the whole body

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20holt Chemistry Section Review Answer

Holt Rinehart & Winston



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Holt Chemistry** Ralph Thomas Myers, 2004 **Principles of Modern Chemistry** David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *The Development of Modern Chemistry* Aaron J. Ihde, 1984-01-01 From ancient Greek theory to the explosive discoveries of the 20th century this authoritative history shows how major chemists their discoveries and political economic and social developments transformed chemistry into a modern science 209 illustrations 14 tables Bibliographies Indices Appendices Holt Chemistry Holt Rinehart & Winston, 2003-01-24 *Modern Chemistry*, 1990 Holt General Science William L. Ramsey, 1988 Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 The Chemical News and Journal of Physical Science, 1870 **Chemical News and Journal of Industrial Science**, 1892

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