



**Advanced Cardiovascular Life Support
Exam Version A (50 questions)**

Please do not mark on this exam. Record the best answer on the separate answer sheet.

- 1. You find an unresponsive patient who is not breathing. After activating the emergency response system, you determine that there is no pulse. What is your next action?**
 - A. Open the airway with a head tilt–chin lift.
 - B. Administer epinephrine at a dose of 1 mg/kg.
 - C. Deliver 2 rescue breaths each over 1 second.
 - D. Start chest compressions at a rate of at least 100/min.**

- 2. You are evaluating a 58-year-old man with chest pain. The blood pressure is 92/50 mm Hg, the heart rate is 92/min, the nonlabored respiratory rate is 14 breaths/min, and the pulse oximetry reading is 97%. What assessment step is most important now?**
 - A. PETCO₂
 - B. Chest x-ray
 - C. Laboratory testing
 - D. Obtaining a 12-lead ECG**

- 3. What is the preferred method of access for epinephrine administration during cardiac arrest in most patients?**
 - A. Intraosseous
 - B. Endotracheal
 - C. Central intravenous
 - D. Peripheral intravenous**

- 4. An activated AED does not promptly analyze the rhythm. What is your next action?**
 - A. Begin chest compressions.**
 - B. Discontinue the resuscitation attempt.
 - C. Check all AED connections and reanalyze.
 - D. Rotate AED electrodes to an alternate position.

Aha Acls Answer Key

Siddappa N.Byrareddy



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Medicine Question-Answer Mr. Rohit Manglik, 2024-07-30 A collection of clinically oriented questions and answers for medical students to test their knowledge and prepare for competitive exams Mastering ACLS, 2006 Mastering ACLS Second Edition will help health care professionals successfully complete the American Heart Association's Advanced Cardiac Life Support course and certification exam This quick review outline format study guide covers the core knowledge and skills needed for ACLS including detailed illustrated life support techniques core case studies arrhythmias defibrillation procedures I V techniques cardiovascular pharmacology myocardial infarction and special resuscitation situations Now in full color this edition features 200 illustrations and algorithms expanded chapter quizzes a comprehensive test expanded information on automated external defibrillators and additional case scenarios in the Megacode and other chapters A new appendix reviews Basic Cardiac Life Support guidelines **ACLS (Advanced Cardiac Life Support) Review: Pearls of Wisdom, Third Edition** Michael Zevitz, Scott Plantz, William Gossman, 2007-11-28 MAXIMIZE YOUR ACLS SCORE A Doody's Core Title 3 STAR DOODY'S REVIEW This book's most impressive aspect is that it presents the core topics in a direct and concise manner that makes it a quick read and easy to understand Doody's Review Service Pearls of Wisdom ACLS Review 3e is your most effective weapon in preparing for the American Heart Association's Advanced Cardiac Life Support exam This rapid effective review ensures your best possible performance on the exam in the shortest possible study time There is no easier or faster way to ensure that you perform your best FEATURES 1200 rapid fire questions cover all topics on the ACLS exam Powered with performance improving pearls and other potent learning aids Updated to reflect the 2005 AMA guidelines NEW chapter on child and infant CPR and ventilation *How to Survive ACLS!* David P. Doernbach, 1998 Critical Care Challenges provides comprehensive up to date information to help nurses keep pace with the latest technological advances newest treatments most current diagnostic tests proven standards of care and evidence based nursing interventions Useful to nurses new to critical care as well as experienced critical care nurses who need more in depth information the book covers pathophysiology clinical assessment diagnosis medical treatments procedures and nursing interventions Body system organization and repetitive entry formats make locating essential information fast and easy The book contains entries on more than 125 disorders treatments and procedures ACLS Review Made Incredibly Easy Lippincott, 2012-03-14 ACLS Review Made Incredibly Easy 2nd edition covers ACLS fundamentals including detailed illustrated information on techniques for life support real life case scenarios arrhythmia recognition and treatment defibrillation procedures I V techniques cardiovascular pharmacology myocardial infarction recognition and treatment special resuscitation situations and more Features such as an easy to use bulleted format that contains illustrations algorithms and graphics help clarify ACLS guidelines Quick quizzes and a comprehensive test help the student put ACLS knowledge to the test ACLS Review Made Incredibly Easy 2nd edition will help health care professionals successfully

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of

naoh naoh

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{NaOH} \cdot \text{Na}_2\text{C}_2\text{O}_4 \cdot 4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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 $\text{NaOH (aq)} + \text{C}_6\text{H}_4\text{COOH (aq)} \rightarrow \text{C}_6\text{H}_4\text{COONa (aq)} + \text{H}_2\text{O (l)}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the

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