#### Stoichiometry Worksheet # 1

 Glucose is used as a source of energy by the human body. The overall reaction in the body is C6H12O6+6O2 → 6CO2 + 6H2O

Calculate the number of moles of oxygen needed to oxidize 12.5 g of C6H12O6.

Ammonia is synthesized from hydrogen and nitrogen according to the following equation.

$$N_2 + 3H_2 \rightarrow 2NH_3$$

If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH<sub>3</sub>) can be produced?

3) Assume that in the decomposition of potassium chlorate, KCLO<sub>3</sub>,

80.5 moles of O2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,

9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" PtCl<sub>2</sub>(NH<sub>2</sub>)<sub>2</sub>, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction

$$K_2PtCl_4 + 2NH_3 \longrightarrow 2KCl + PtCl_2(NH_3)_2$$

- A. How many moles of "cisplatin" can be produced from 2.50 g K2PtCl4?
- B. How many moles of NH3 would be needed if 3.8 moles of K2PtCl4 were present?

# **Answer Key To Stoichiometry Homework Problems**

**LL Leslie** 

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In a global used by monitors and the ceaseless chatter of instantaneous transmission, the melodic splendor and mental symphony developed by the prepared term often disappear into the back ground, eclipsed by the constant noise and distractions that permeate our lives. But, set within the pages of **Answer Key To Stoichiometry Homework Problems** a marvelous literary treasure brimming with fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, that interesting masterpiece conducts visitors on a mental trip, skillfully unraveling the hidden melodies and profound impact resonating within each cautiously crafted phrase. Within the depths with this emotional assessment, we can discover the book is central harmonies, analyze its enthralling writing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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