# **Buffers Keep the Balance**

ata Table 1. Model Carbonate Blood Buffer

Table 1	Test Tube Number						
	1	2	3	4	5	6	
Solution	Carbonic Acid (Reference)	Model Carbonate Blood Buffer	Water (Control)	Model Carbonate Blood Buffer	Water (Control)	Sodium Bicarbonate (Reference)	
Bromthymol blue indicator color	Yenow .	green 8h	green 600	men 13h	9 Heen 600	bive	
Initial pH value	6.0	7.2	6.0	7.0	6.0	8.0	
Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color)	NA	16	5	NA	NA	NA	
Number of drops of NaOH required to convert solutions 4 and 5 to blue "hase reference color)	NA	NA	NA	2	2	NA	

# Data Table 2. Effect of HCl on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color				
		Initial	After 1 drop of HCl	After 5 drops of HCI	After 10 drops of HCl	
Water (control)	6.0	6.0	6.0	NA	NA	
Phosphate Buffer A	6.0	60	6.0	6.0		
Phosphate Buffer B	6.4	6.4	6.4	6.0	6.0	

# Data Table 3. Effect of NaOH on Biological Phosphate Buffers

		Universal Indicator Color				
	(pH paper)	Initial	After 1 drop of NaOH	After 5 drops of NaOH	After 10 drops of HCI	
(control)	6.0	6.0	7.0	NA	NA NA	
Phosphate Buffer A	6.0	6.0	6.8	9 .	200	
Phosphate Buffer B	6.4	6.4	8.0	9.0	8.0	

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**Marco Cascella** 

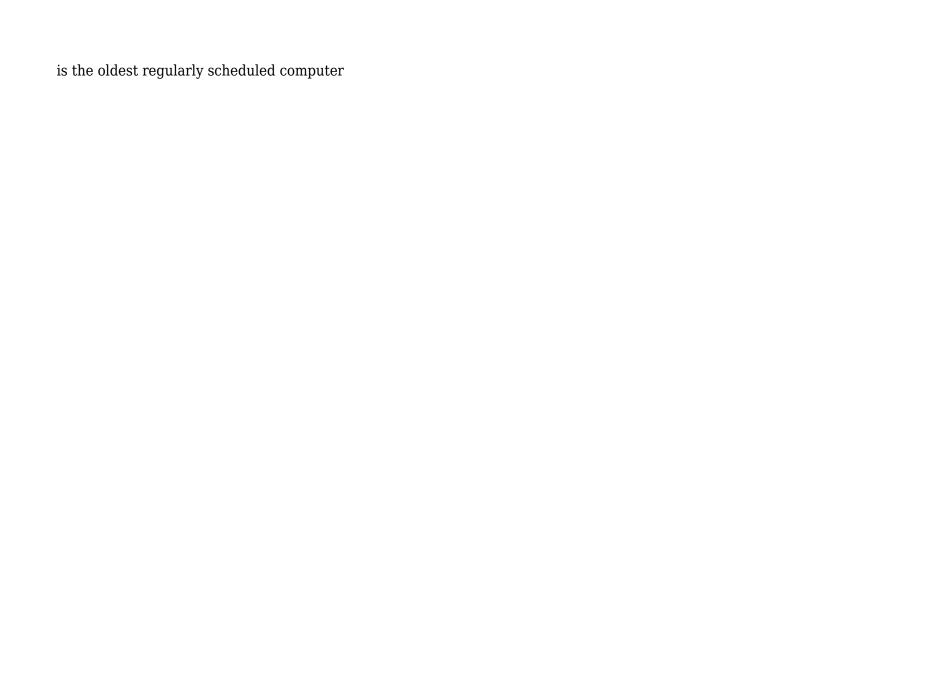
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