

Loyola School
Second Model Examination
CHEMISTRY



[15]

Name Albanard M. Nair

Div A Roll No. 1

SCIENCE Paper - 2

(Two hours)

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

The intended marks for questions or parts of questions are given in brackets []

SECTION I (40 Marks)

(Attempt ALL questions)

Question 1

Choose the correct answer from the options given below:

(Do not copy the question; write the correct answers only.)

 Whenever the gases react under similar conditions of temperature and pressure, they do so in volumes which bear a simple ratio of each other and the products if gaseous. The above law was stated by

a) Boyle b) Charles c) Gay Lussac d) Avogadro

A strip of copper is placed in four different colourless salt solutions. After electrolysis, one of the salt solutions becomes blue in colour. Identify the salt solution which turns blue from the given salt solutions:

a) K₂SO₄ b) CuSO₄ c) AgSO₄ d) CaSO₄

Potassium is the element below sodium in the long form of the periodic table. The electronegativity of potassium will be

a) Lower than sodium b) Higher than sodium c)Same as sodium d) None of these

4. The salt solution which does not react with ammonium hydroxide is :

a) Calcium Nitrate b) Zinc Nitrate c) Lead Nitrate d) Copper Nitrate

3. Assertion: Sodium is the most reactive metal among group 1 elements.

Reason: In case of metals, greater the tendency to lose electrons, greater is the reactivity.

a) A and R both are correct statements and R is the correct explanation of the A

b) A and R both are correct statements but R is not the correct explanation of the A

c) A is true, but R is false

d) A is false, but R is true

Answers For A Ls Chemistry

A Gutmann

Answers For A Ls Chemistry:

Unveiling the Magic of Words: A Overview of "Answers For A Ls Chemistry"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Answers For A Ls Chemistry**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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