Name of Street, Street

Covalent Bonding Worksheet Covalent bonding occurs when two or more NON METALS share electrons, attempting

to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will

be shared between the elements.

1.) He traderspore in distribution Single Bond  $H + H \longrightarrow H \longrightarrow H - H$ 

2.) F2 thurstone is distributed F F F F F F F

3.) O₂ oxygon is distrossis
O + O → O=O Double Bond

4.) N2 mitrogen is distornic. In this a triple bond?

N + N

5.) BF3 you need 3 flooring atoms here

B + F

6.) Ammonia NH3 hint how many hydrogen atoms are needed?

N + H

7.) Carbon dioxide CO2

O + C + O

Methane CH<sub>4</sub> careful here 4 hydrogen atoms needed

C + H

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

H + O + H

10.) SO<sub>2</sub> hint: one pair of electrons from sulfur must be slit up for this one to work.

0 + s + 0

# **Answer Key To The Nature Of Covalent Bonding**

**RD Boyd** 

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