

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? *Frank can share his red marker with Flora and Flora can share her blue marker with Frank.*
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? *Olivia can share her purple and green markers with Otto and Otto can share his black and brown markers with Olivia.*

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bond's Gizmo*, atoms form bonds in this way.

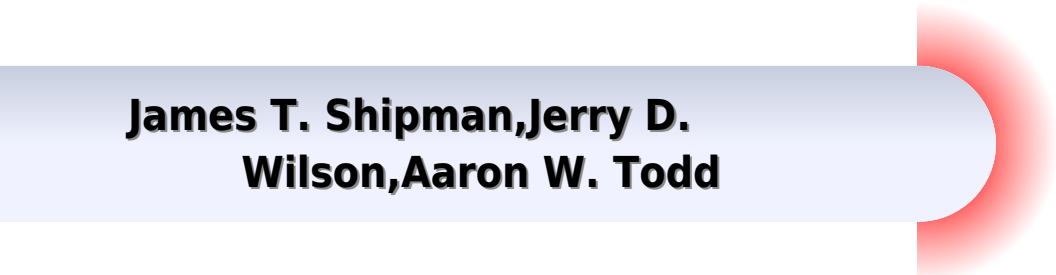


To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.

1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? *Each fluorine atom has seven valence electrons.*
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens? *The electrons that was dragged now orbits both atoms.*
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now? *A pair of electrons now orbit both atoms.*

Answer Key To Student Exploration Covalent Bonds

**James T. Shipman, Jerry D.
Wilson, Aaron W. Todd**



Answer Key To Student Exploration Covalent Bonds:

Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Sengupta Printed by IECIT Printing and Publication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials Prentice Hall Chemistry ,2000 Prentice Hall Science Explorer Michael J. Padilla,2002

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The Software Encyclopedia 2001, 2001 *Holt Chemistry* Ralph Thomas Myers, 2004 *The Art and Science of Questions* Ronald Legarski, 2024-09-28 In *The Art and Science of Questions* Ronald Legarski explores the profound power of questioning in shaping human thought innovation and discovery Whether in business science education or everyday life the ability to ask the right question can lead to deeper understanding better decisions and groundbreaking ideas This book offers a comprehensive guide to mastering the art of inquiry covering various types of questions from open ended to leading reflective and hypothetical and providing practical methods for applying them in real world contexts With insights drawn from philosophy psychology leadership and technology this book equips readers with the tools to ask more effective questions and unlock the full potential of inquiry Legarski also delves into the future of questioning examining how artificial intelligence virtual reality and quantum computing will revolutionize how we ask and answer questions in the coming decades Whether you're a student a professional or simply a curious mind *The Art and Science of Questions* will transform the way you approach problem solving learning and communication

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Biology Neil A. Campbell, Jane B. Reece, 2005

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