

Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine the structure of the atom by observing the emission spectrum of hydrogen.

Part I: Finding Evidence for Discrete Energy Levels

When hydrogen gas is excited, it emits light. The light is composed of several discrete lines of color. The lines are called the Balmer series. The lines are named after the physicist who discovered them. The lines are shown in the diagram below.

Observation: The emission spectrum of hydrogen consists of several discrete lines of color. The lines are named after the physicist who discovered them. The lines are shown in the diagram below.

From the observation of the emission spectrum of hydrogen, we can conclude that the energy levels of the atom are discrete.

Conclusion: The emission spectrum of hydrogen consists of several discrete lines of color. The lines are named after the physicist who discovered them. The lines are shown in the diagram below.

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Observation: The emission spectrum of hydrogen consists of several discrete lines of color. The lines are named after the physicist who discovered them. The lines are shown in the diagram below.

From the observation of the emission spectrum of hydrogen, we can conclude that the energy levels of the atom are discrete.

Answers Lesson 11 Atomic Pudding

PT Brinkman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the PT Brinkman text.

Answers Lesson 11 Atomic Pudding:

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