

Oxidation-Reduction Worksheet

For each reaction below, identify the atom oxidized, the atom reduced, the oxidizing agent, the reducing agent, the oxidation half reaction, the reduction half reaction, and then balance the equation by the method of oxidation-reduction showing all electrons transfers.

1. $\overset{0}{\text{Mg}} + 2\overset{+1}{\text{H}}\overset{-1}{\text{Cl}} \rightarrow \overset{+2}{\text{Mg}}\overset{-2}{\text{Cl}}_2 + \overset{0}{\text{H}}_2$ $\overset{0}{\text{Mg}} \xrightarrow{-2e^-} \overset{+2}{\text{Mg}}$ $2\overset{+1}{\text{H}} \xrightarrow{+2e^-} \overset{0}{\text{H}}_2$
2. $2\overset{0}{\text{Fe}} + 3\overset{+3}{\text{V}}_2\overset{-2}{\text{O}}_3 \rightarrow \overset{+3}{\text{Fe}}_2\overset{-2}{\text{O}}_3 + 6\overset{+2}{\text{V}}\overset{-2}{\text{O}}$ $2\overset{0}{\text{Fe}} \xrightarrow{-6e^-} \overset{+3}{\text{Fe}}_2$ $3\overset{+3}{\text{V}}_2 \xrightarrow{+6e^-} 6\overset{+2}{\text{V}}$
3. $\overset{+7}{\text{K}}\overset{+4}{\text{Mn}}\overset{-2}{\text{O}}_4 + \overset{+5}{\text{K}}\overset{-2}{\text{N}}\overset{-2}{\text{O}}_2 + 2\overset{+2}{\text{H}}_2\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4 \rightarrow \overset{+2}{\text{Mn}}\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4 + 2\overset{+2}{\text{H}}_2\overset{-2}{\text{O}} + \overset{+5}{\text{K}}\overset{-2}{\text{N}}\overset{-2}{\text{O}}_3 + \overset{+6}{\text{K}}_2\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4$
4. $\text{K}_2\text{Cr}_2\text{O}_7 + \text{SnCl}_2 + \text{HCl} \rightarrow \text{CrCl}_3 + \text{SnCl}_4 + \text{KCl} + \text{H}_2\text{O}$
5. $\text{KMnO}_4 + \text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{Cl}_2 + \text{K}_2\text{SO}_4 + \text{MnSO}_4 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$
6. $\text{K}_2\text{Cr}_2\text{O}_7 + \text{H}_2\text{O} + \text{S} \rightarrow \text{SO}_2 + \text{KOH} + \text{Cr}_2\text{O}_3$
7. $\text{KClO}_3 + \text{C}_{12}\text{H}_{22}\text{O}_{11} \rightarrow \text{KCl} + \text{H}_2\text{O} + \text{CO}_2$
8. $\text{H}_2\text{C}_2\text{O}_4 + \text{K}_2\text{MnO}_4 \rightarrow \text{CO}_2 + \text{K}_2\text{O} + \text{Mn}_2\text{O}_3 + \text{H}_2\text{O}$
9. $\text{Mn}(\text{NO}_3)_2 + \text{NaBiO}_3 + \text{HNO}_3 \rightarrow \text{HMnO}_4 + \text{Bi}(\text{NO}_3)_3 + \text{NaNO}_3 + \text{H}_2\text{O}$
10. $\text{H}_2\text{C}_2\text{O}_4 + \text{KMnO}_4 \rightarrow \text{CO}_2 + \text{K}_2\text{O} + \text{Mn}_2\text{O}_3 + \text{H}_2\text{O}$

Analyzing Oxidation Reduction Reactions Answer Key

Arshad Iqbal



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Publications of the National Bureau of Standards ... Catalog United States. National Bureau of Standards, 1978

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Methyl Furfural HMF Purified Terephthalic Acid Polyethylene 2,5-Furandicarboxylate MEG *Polyethylene Furanoate Production from Raw Sugar - Cost Analysis - PEF E31B* Intratec, 2019-09-17 This report presents a cost analysis of Polyethylene Furanoate (PEF) production starting from raw sugar and monoethylene glycol (MEG). The process examined is similar to the Avantium YXY process followed by a process that copolymerizes FDCA and MEG. In this process, a raw sugar sucrose is inverted to glucose and fructose. The glucose is isomerized to fructose. Then fructose is converted to methoxy methyl furan (MMF) intermediate which is oxidized to FDCA. This process generates methyl acetate and methyl levulinate as by-products. The FDCA generated and MEG are polymerized to PEF in two polymerization steps in order to obtain Bottle Grade PEF. This report was developed based essentially on the following references: 1. US Patent 8519167 issued to Furanix Technologies B.V. in 2013; 2. US Patent 20160024039 issued to Furanix Technologies B.V. in 2016; 3. WO Patent 2015137807 issued to Furanix Technologies in 2015. Keywords: Polyethylene Furanoate (PEF), Green FDCA, Hydroxyl Methyl Furfural (HMF), Purified Terephthalic Acid, Polyethylene 2,5-Furandicarboxylate (MEG).

Handbook of Water Analysis Leo M.L. Nollet, Leen S. P. De Gelder, 2000-06-27 This work details water sampling and preservation methods by enumerating the different ways to measure physical, chemical, organoleptical, and radiological characteristics. It provides step-by-step descriptions of separation, residue determination, and cleanup techniques for a variety of fresh and salt waters. It also discusses information regarding the analysis and detection of bacteria and algae.

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