

Abstract

Type of spectrum	Photographic example
Continuous (or continuum)	
Absorption (dark line)	
Emission (bright line)	

A continuous visible spectrum showing colors from violet (A) to red (A). The spectrum is labeled with letters A through H from left to right. Below the spectrum is a scale for wavelength in nm, ranging from 400 to 700 nm.

Designation	Element	Wavelength (nm)	Designation	Element	Wavelength (nm)
F	O ₂	844.665	a	Fe	486.265
E	O ₂	844.694	F	H β	486.134
A	O ₂	779.370	d	Fe	466.814
B	O ₂	684.739	e	Fe	438.333
C	H α	656.285	G'	H γ	434.047
e	O ₂	627.661	G	Fe	430.790
D ₁	Na	589.592	G	Ca	430.774
D ₂	Na	589.093	h	H δ	410.173
D ₂ or d	H α	587.5628	H	Ca ⁺	396.847
e	H β	448.013	K	Ca ⁺	393.368
F ₁	Fe	527.009	L	Fe	382.044
h ₁	Mg	518.362	N	Fe	358.123
h ₂	Mg	527.276	P	Ti	336.132
h ₃	Fe	516.891	T	Fe	302.108
h ₄	Fe	516.892	t	Si	299.444
h ₅	Mg	516.733			

-

[illegible]

5. A Nebula of mainly Hydrogen and Mercury (unlikely) is between a star and the earth.
- a. What happens to the light as it passes through the nebula
- _____
- _____
- b. The Star is mainly Hydrogen and Helium. Complete the spectrum



Analysis Of Spectral Lines Answers

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Analysis Of Spectral Lines Answers:

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Stedman,1872 Chemical News and Journal of Industrial Science ,1890
Lil-'Ulūm Al-Riyāḍīyah Wa-al-Ṭabī'īyah ,1946

Majmū'at Abḥāth Al-Jam'īyah Al-Miṣrīyah

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