

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

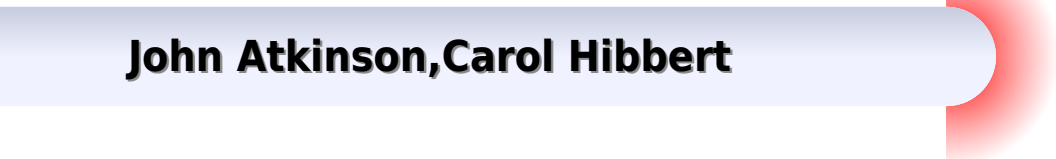
	Acid/Base?	Name
HNO_3	A	nitric acid
$Mg(OH)_2$	B	magnesium hydroxide
H_3PO_4	A	phosphoric acid
H_2CO_3	A	carbonic acid
HF	A	hydrofluoric acid
$Fe(OH)_3$	B	iron (iii) hydroxide

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

1 Answers Acids And Bases

John Atkinson, Carol Hibbert



1 Answers Acids And Bases:

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Fluid, Electrolyte, and Acid-Base Imbalances Allison Hale, Mary Jo Hovey, 2013-10-29 Here's all of the crucial coverage you need to succeed in class and confidently prepare for your classroom exams and the NCLEX. Easy to follow outlines focus on the information essential to make this challenging subject more manageable.

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