

UNDERSTANDING GRADE BOUNDARIES

Grade	Description
9	Exceptional achievement
8	High achievement
7	Good achievement
6	Satisfactory achievement
5	Strong pass
4	Standard pass

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calorimeter with a total heat capacity of 15 48 kJ $^{\circ}\text{C}^{-1}$ the temperature of the calorimeter and its contents changes from 19 77 $^{\circ}\text{C}$
to 33 90 $^{\circ}\text{C}$ calculate ΔH for the following reaction as written $2\text{NH}_3(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g}) + 2\text{CH}_4(\text{g})$

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final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the
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water is 4 184 J $\text{g}^{-1} \text{ } ^{\circ}\text{C}^{-1}$ $m = 5 650 \text{ g}$ H_2O 4 184 J $\text{g}^{-1} \text{ } ^{\circ}\text{C}^{-1}$ 55 4 $^{\circ}\text{C}$ 1310 kJ

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