

14

THE BEHAVIOR OF GASES

Chapter Quiz

Indicate whether the statement is true or false.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to its pressure.
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to its temperature in kelvins.
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to its temperature in kelvins.
4. The combined gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.
5. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.
6. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

Indicate whether the statement is true or false.

7. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

Indicate whether the statement is true or false.

8. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

Indicate whether the statement is true or false.

9. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to its pressure.	True
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to its temperature in kelvins.	True
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to its temperature in kelvins.	True
4. The combined gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.	True
5. The ideal gas law states that, for a fixed amount of gas, the product of the pressure and volume is proportional to the absolute temperature.	True

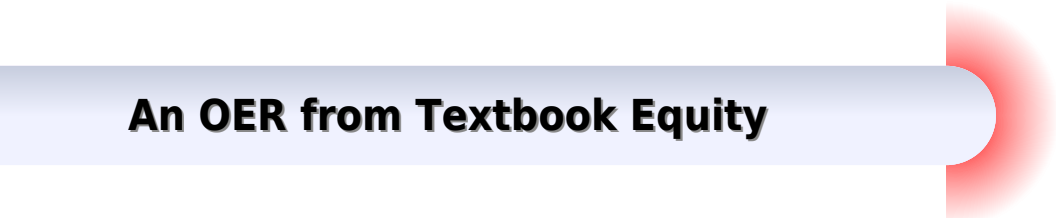
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 - Chapter 4: 1the Behavior Of Gases Chapter Quiz in Specific Contexts
 - Chapter 5: Conclusion
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