

Analytical Chemistry
3rd Long Exam

PART I. MULTIPLE CHOICE. Choose the letter that corresponds to the **BEST** answer. Write **[]** when the answer is **NOT** in the choices. (25 pts)

1. Transfer of electrons from one reactant to another

- a. precipitation rxn c. redox rxn
b. half rxn d. neutralization rxn

2. What is the mole ratio of $\text{Ce}(\text{IV})$ to NO_2^- for the oxidation of nitrite to nitrate using Ce^{4+} in the presence of an acid:



- a. 1:2 c. 1:1
b. 2:1 d. 2:2

3. How many mole(s) of electron is(are) produced in the oxidation of sodium bisulfate (NaHSO_4) to BiO^+ and H^+ under acidic conditions?

- a. 1 c. 3
b. 2 d. 4

4. Azide (N_3^-) was added in the determination of the dissolved oxygen in water sample to remove interferences caused by the presence of what ion?

- a. chloride c. nitrite
b. fluoride d. nitrate

5. Which of the following statements is(are) true?

- I. Prepared KMnO_4 solutions should be stored in amber bottles
II. KMnO_4 solutions are boiled to remove CO_2 which makes the solution basic
III. $\text{Mn}(\text{II})$ is the product of the degradation of KMnO_4
IV. KMnO_4 solutions are filtered to remove manganese(IV) oxide

- a. I only c. I and IV
b. I and II d. I, II and III

6. What is the oxidation state of carbon in oxalic acid?

- a. +2 c. +6
b. +4 d. +7

7. The complex formed between $\text{Cu}(\text{I})$ and 1,10-phenanthroline has a molar absorptivity of $7000 \text{ L cm}^{-1} \text{ mol}^{-1}$ at 435 nm, the wavelength of maximum absorption. What is the absorbance of a $5.17 \times 10^{-3} \text{ M}$ solution of the complex when measured in a 1.00-cm cell at 435 nm?

- a. 0.0431 c. 0.432
b. 0.0432 d. 0.431

8. In the presence of iodine, thiosulfate ion is oxidized quantitatively to tetrathionate ion. What is the chemical formula of tetrathionate ion?

- a. $\text{S}_4\text{O}_7^{2-}$ c. $\text{S}_4\text{O}_6^{2-}$
b. $\text{S}_4\text{O}_5^{2-}$ d. $\text{S}_4\text{O}_8^{2-}$

9. In Winkler situation, the flocc is produced from

- a. oxidation of I^- to I_2 c. reduction of I^- to I_2
b. oxidation of $\text{Mn}(\text{II})$ to $\text{Mn}(\text{III})$ d. reduction of $\text{Mn}(\text{II})$ to $\text{Mn}(\text{III})$

10. A component of a UV-Vis spectrophotometer that restricts the radiation being measured to a narrow band that is absorbed or emitted by the analyte.

- a. light source c. radiation filter
b. detector d. wavelength selector

11. Component of a UV-Vis instrument that operates by blocking or absorbing all but a restricted band of radiation.

- a. light source c. radiation filter
b. detector d. wavelength selector

12. Oxidation of VO^{2+} using permanganate or cerium as standard solution produces $\text{VO}(\text{OH})_2^{+}$, what is the oxidation state of vanadium in $\text{VO}(\text{OH})_2^{+}$?

Analytical Chemistry Exam Questions

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Analytical Chemistry Exam Questions:

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Principles and Practice of Analytical Chemistry F. W. Fifiield,David Kealey,2000-09-05 Over more than two decades this book has established itself as the first choice for growing numbers of students and practising analysts who require a well written and concise overview of the principles and practice of analytical chemistry Recurring themes are improvement in medicine and the environment the I T revolution and its continuing impacts on both analytical methodology and data handling The successive editions of Principles and Practice of Analytical Chemistry have kept pace with the associated

developments in the subject Significant changes encompass strengthening of the coverage of data handling together with new material covering rapidly developing subject areas of molecular spectrometry atomic spectrometry and separation techniques Answers to the self learning problems and exercises are also included All sections of the book have been fully reviewed and updated as appropriate The new edition of the book continues to provide a sound and broad base for the study of analytical chemistry by undergraduate and postgraduate students and to be a useful resource for practising analysts seeking a summary of the principles of techniques and methods

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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the Berri Lambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

TRAC: Trends in Analytical Chemistry U A Th Brinkman, J. G. Dorsey, J R Durig, 2013-09-24 TRAC Trends in Analytical Chemistry Volume 7 provides information pertinent to the trends in the field of analytical chemistry This book discusses a variety of topics related to analytical chemistry including biomolecular mass spectroscopy affinity chromatography electrochemical detection nucleosides and protein sequencing Organized into 63 parts encompassing 158 chapters this volume begins with an overview of the significance of quality and productivity in the analytical laboratory This text then presents a comprehensive review on alcohol dehydrogenases immobilization and applications in analysis and synthesis Other chapters consider the various tests

for determining the excellence of quantitative assays available for analysts to utilize for method validation This book discusses as well the primary challenge of neuropharmacologists to relate physiological functions to the many ligand binding sites identified in brain tissue The final chapter deals with the fundamentals and applications of biosensors This book is a valuable resource for analytical chemists chemical engineers clinical chemists neuropharmacologists and scientists

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Analytical Chemistry I Ulf Ritgen, 2023-04-21 This workbook takes you through the successful work Harris Textbook of Quantitative Analysis and is designed primarily for self study In five parts the lecture content of analytical chemistry is summarized and explained using selected examples Basic concepts of analytical chemistry are presented as well as the principle and various techniques of dimensional analysis and chromatography UV VIS infrared and Raman spectroscopy are used to explain the investigation of molecularly present compounds and selected techniques of atomic spectroscopy conclude the introduction to the fundamentals of analysis The textbook's essential sections and illustrations are repeatedly referred to which facilitates independent learning of the fundamentals of analytical chemistry Easy to read the book introduces the fundamentals and key techniques of analytical chemistry it is aimed at undergraduate students of chemistry or related science subjects It repeatedly refers back to the basics familiar from courses in general chemistry so that the connections between what is already known and what is new become immediately apparent Learning with this workbook has been tested in a distance learning chemistry course and facilitates preparation for module examinations in analytical chemistry This book is a translation of the original German 1st edition *Analytische Chemie I* by Ulf Ritgen published by Springer Verlag GmbH Germany part of Springer Nature in 2019 The translation was done with the help of artificial intelligence machine translation by the service DeepL.com A subsequent human revision was done primarily in terms of content so that the book will read stylistically differently from a conventional translation Springer Nature works continuously to further the development of tools for the production of books and on the related technologies to support the authors

Chemical Analysis for Forensic Evidence Arian van Asten, 2022-11-24 Chemical Analysis for Forensic Evidence provides readers with the fundamental framework of forensic analytical chemistry describing the entire process from crime scene investigation to evidence sampling laboratory analysis quality aspects and

reporting and testifying in court In doing so important principles and aspects are demonstrated through the various forensic expertise areas in which analytical chemistry plays a key role including illicit drugs explosives toxicology fire debris analysis and microtraces such as gunshot residues glass and fibers This book illuminates the underlying practical framework that governs how analytical chemistry is used in practice by forensic experts to solve crime Arian van Asten utilizes a hands on approach with numerous questions examples exercises and illustrations to help solidify key concepts and teach them in an engaging way Provides a forensic analytical chemistry framework based on how professionals actually use chemistry to solve crimes Introduces leading principles necessary to forensic practice understanding Answers key questions with a wealth of illustrations and real world examples

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