# Anions, Cations, and Ionic Reactions

Hands-On Labs, Inc. Version 42-0304-00-01

#### Lab Report Assistant

This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report. The intent is to facilitate students' writing of lab reports by providing this information in an editable file which can be sent to an instructor.

## **Exercise 1: Qualitative Anion Tests**

Data Table 1. Anion Confirmation Tests.

Chemical	Anion	Observations: Addition of AgNO <sub>3</sub>	Observations: Addition of HCI
NaBr	Bromide	Turned milky white	Turned yellowish More of a gel
Na <sub>2</sub> CO <sub>3</sub>	Carbonate	Thick white glob A little grainy	White sediment in a clear liquid Bubbles up
NaCl	Chloride	Milky white	White sediment visible
Nal	lodide	Yellow glob immediately when dropped in	Yellow, grainy/sediment, thick looking
Na <sub>2</sub> SO <sub>4</sub>	Sulfate	Thickened	Very cloudy/milky
Na₂S	Sulfide	Brown Globs immediately when dropped in Still watery	Brown globs thinned out into brown pieces of sediment Still watery (brownish) looks like dirty water
Unknown #103	Carbonate	Thick white glob Turned gel	Bubbly and cloudy Very thick gel
Unknown #104	Sulfide	Brown globs immediately when dropped in Still watery	Brown globs thinned out into brown pieces of sediment Still watery (brownish) looks like dirty water

# **Analysis Of Anions Cations Lab Answers**

Water Resources Scientific Information Center

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