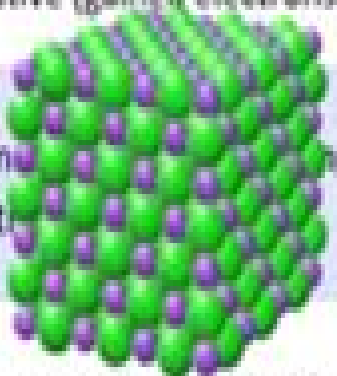


Atoms, Ions, & Molecules

	Atom		Ion		Molecule
Number of particles		One 	One or more 	 	Two or more 
Charge	Neutral (equal protons and electrons)		Positive (lost electrons) negative (gained electrons)		Neutral (electrons shared between atoms)
Formation	Cannot be further broken down without losing chemical identity.		Formed by losing or gaining electrons 		Formed by sharing electrons between atoms.
Examples	Helium (He), Oxygen (O), Iron (Fe)		Sodium ion (Na ⁺), Chloride ion (Cl ⁻), Sulfate ion (SO ₄ ²⁻)		Water (H ₂ O), Carbon dioxide (CO ₂), Sugar (C ₆ H ₁₂ O ₆)



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Chemistry John Olmsted, Gregory M. Williams, 1997 Textbook outlining concepts of molecular science

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