ENANCE	

(Months or adometer entering whichever accurs first)	months	6	12	18	24	30	36	42	48	54	60	66	72	78	84	90	96	102	108	114	120
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Engine	1			-					7			-					-				
Drive belts	6 4	2	4		1		1		1.		R		1		113		1		1		R
Engine oil.		R	R	R	R	R	R	R	R	R	R	R	R	R	8	R	R	R	R	R	R
Engine oil filter		R	R	R	R	R	R	R	R	R	8	R	R	R	R	R	R	R	Я	R	R
Cooling and heater system	13	1	1		7				1		1		1		11		-1		1		1
Engine coolant - Include i coolant (Toyota Genuine 5		1			h				1				1				R				1
Exhaust pipes and mounti	ngs		11	1	0	N	1		1		1		1				1.		1		1
Electrical System				1		8															
Spark plugs					1	10					8							7,0			R
Battery		1	1.	1.	1	m	3	-	1	-1	1	1	1	1	1	1	- 1	1	1	1	-1
Fuel System						No	30	4													
Fuel filter, including in-tan	k filter						Æ	1	4				R								
Engine air cleaner filter			1		R	10.0	P	4	R	١.	-		R		4		R		1		R
Fuel tank cap, fuel lines, or and fuel vapor control valv					1			1	Y		1		Ī		i.		i		1		i,
Charcoal canister					1				A	1	1	1	al.				1				1

Inspect the condition of the radiator and condenser. Check that they are not damaged or blocked with leaves, dirt, or insects. Inspect the hoses and connections for installation condition and corrosion.

Replace the engine coolant initially at 96 months or 160,000 km (whichever occurs first), then every 48 months or 80,000 km (whichever occurs first). Use Genuine Toyota Super Long Life Coolant (SLLC) or a similar high guality ethylene glycol based non-slicate, non-amine, non-nitrite, and non-borate coolant with long-life hybrid organic acid technology. (Coolant with long-life hybrid organic acid technology consists of the combination of low phosphates and organic acids.)

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web part a aqueous reactions simulation your instructor will play a simulation for you during lab and guide you through the questions in the first part of this assignment you should answer these questions based on what you observe in the simulation be as specific as possible in your answers

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web students review precipitation reactions and use the virtual lab to find out which chemicals react to form a precipitate using stoichiometry to connect mass to concentration students determine if the water is safe to drink based on epa guidelines