

1999 Advanced Placement Chemistry Exam

Section I: Multiple Choice

Time — 1 hour and 30 minutes

NO CALCULATORS MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

T = temperature
 P = pressure
 V = volume
 S = entropy
 H = enthalpy
 G = free energy
 R = molar gas constant
 n = number of moles
 M = molar

m = molal
 L, mL = liter(s), milliliter(s)
 g = gram(s)
 nm = nanometer(s)
 atm = atmosphere(s)
 J, kJ = joule(s), kilojoule(s)
 V = volt(s)
 mol = mole(s)

Part A

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1–4 refer to the following types of energy.

- (A) Activation energy
- (B) Free energy
- (C) Ionization energy
- (D) Kinetic energy
- (E) Lattice energy

1. The energy required to convert a ground-state atom in the gas phase to a gaseous positive ion
2. The energy change that occurs in the conversion of an ionic solid to widely separated gaseous ions
3. The energy in a chemical or physical change that is available to do useful work
4. The energy required to form the transition state in a chemical reaction

Questions 5–8 refer to atoms for which the occupied atomic orbitals are shown below.

- (A) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (B) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (C) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (D) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (E) $[Ar] 4s \underline{\hspace{1cm}} 3d \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$

5. Represents an atom that is chemically unreactive
6. Represents an atom in an excited state
7. Represents an atom that has four valence electrons
8. Represents an atom of a transition metal

Questions 9–12 refer to aqueous solutions containing 1:1 mole ratios of the following pairs of substances. Assume all concentrations are 1 M .

2008 Ap Chemistry Exam Multiple Choice Answers

Princeton Review (Firm)



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5 Steps to a 5 AP Chemistry, 2008-2009 Edition John T. Moore, Richard H. Langley, 2008-01-04 A PERFECT PLAN FOR THE PERFECT SCORE We want you to succeed on your AP exam That s why we ve created this 5 step plan to help you study more effectively use your preparation time wisely and get your best score This easy to follow guide offers you a complete review of your AP course strategies to give you the edge on test day and plenty of practice with AP style test questions You ll sharpen your subject knowledge strengthen your thinking skills and build your test taking confidence with Full length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules so you can pick the one that meets your needs The 5 Step Plan helps you get the most out of your study time Step 1 Set Up Your Study Program Step 2 Determine Your Readiness Step 3 Develop the Strategies Step 4 Review the Knowledge Step 5 Build Your Confidence **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam **Guide to College Majors 2008** Princeton Review (Firm), 2008-02-05 Fully revised and updated for 2008 this resource includes the most current information available on more than 350 of the most popular majors Each profile

includes an overview of the major career options and earning potential sample college curricula and more **SAGE Handbook of Research on Classroom Assessment** James H. McMillan, 2013 The Sage Handbook of Research on Classroom Assessment provides scholars professors graduate students and other researchers and policy makers in the organizations agencies testing companies and school districts with a comprehensive source of research on all aspects of K 12 classroom assessment The handbook emphasizes theory conceptual frameworks and all varieties of research quantitative qualitative mixed methods to provide an in depth understanding of the knowledge base in each area of classroom assessment and how to conduct inquiry in the area It presents classroom assessment research to convey in depth the state of knowledge and understanding that is represented by the research with particular emphasis on how classroom assessment practices affect student achievement and teacher behavior Editor James H McMillan and five Associate Editors bring the best thinking and analysis from leading classroom assessment researchers on the nature of the research making significant contributions to this prominent and hotly debated topic in education *Developing and Validating Test Items* Thomas M. Haladyna, Michael C. Rodriguez, 2013-07-18 Since test items are the building blocks of any test learning how to develop and validate test items has always been critical to the teaching learning process As they grow in importance and use testing programs increasingly supplement the use of selected response multiple choice items with constructed response formats This trend is expected to continue As a result a new item writing book is needed one that provides comprehensive coverage of both types of items and of the validity theory underlying them This book is an outgrowth of the author's previous book *Developing and Validating Multiple Choice Test Items* 3e Haladyna 2004 That book achieved distinction as the leading source of guidance on creating and validating selected response test items Like its predecessor the content of this new book is based on both an extensive review of the literature and on its author's long experience in the testing field It is very timely in this era of burgeoning testing programs especially when these items are delivered in a computer based environment Key features include Comprehensive and Flexible No other book so thoroughly covers the field of test item development and its various applications Focus on Validity Validity the most important consideration in testing is stressed throughout and is based on the Standards for Educational and Psychological Testing currently under revision by AERA APA and NCME Illustrative Examples The book presents various selected and constructed response formats and uses many examples to illustrate correct and incorrect ways of writing items Strategies for training item writers and developing large numbers of items using algorithms and other item generating methods are also presented Based on Theory and Research A comprehensive review and synthesis of existing research runs throughout the book and complements the expertise of its authors **The Publishers Weekly**, 2008 *Guide to College Visits* Princeton Review, 2007-04-10 A practical handbook covers everything needed to plan a successful visit to campuses across the country with information on nearby accommodations for every budget the best and worst times to visit itineraries and campus tours **The Science Teacher**, 2009 **Get It Together for College** College

Board,2008-06-24 A comprehensive guide for getting through the college application process with tips on entrance essays financial aid forms and campus visits **Oswaal JEE Advance 10 Mock Test Papers (Paper-1 & Paper-2) Physics, Chemistry, Mathematics (For 2023 Exam)** Oswaal Editorial Board,2023-02-27 Benefits of book which distinguish it from others 100% Updated With 2022 Paper 1 2 Fully Solved Extensive Practice With 10 Sample Question Papers 100% Exam Readiness With highly Probable SQPs with a success rate of more than 87% in 2022 Crisp Revision With Mind Maps Mnemonics Appendix Valuable Exam Insights With Subjective Trend Analysis *Oswaal JEE Advance 10 Mock Test Papers (Paper-1 & Paper-2) Physics, Chemistry, Mathematics (For 2024 Exam)* Oswaal Editorial Board,2023-07-01 Description of the product 100% Updated with Fully Solved 2023 Papers 1 2 Extensive Practice with 1200 Practice Questions of Paper 1 Paper 2 Crisp Revision with Smart Mind Maps Mnemonics Appendix Valuable Exam Insights with Expert Tips Tricks and Shortcuts to Crack JEE Advanced Concept Clarity with Extensive Explanations of 10 Mock Test Papers 100% Exam Readiness with Chapter wise Analysis 2018 2023 Science John Michels (Journalist),2008 **Who's Who in Science and Engineering 2008-2009** Who's Who Marquis,Marquis Who's Who,2007-12 **McGraw-Hill's SAT, 2010 Edition** Christopher Black,Mark Anestis,2009-06-05 We want to help you succeed on the SAT We ve put all of our proven expertise into McGraw Hill s SAT to make sure you re ready for this difficult exam With this book you ll get essential skill building techniques and strategies developed by a team of renowned test prep tutors Their innovative teaching methods and expert coaching will help you master every question type With McGraw Hill s SAT we ll guide you step by step through your preparation program and give you the tools you need to succeed McGraw Hill s SAT gives you 5 full length practice SATs in the book 1 full length practice SAT test online 20 model essays to show you exactly what the graders are looking for Hundreds of samples questions with explanations Demonstrations of how to apply test taking strategies Diagnostic techniques to help you create your most effective study plan Online help including practice SATs test taking tips and more SAT is a registered trademark of the College Entrance Examination Board which was not involved in the production of and does not endorse this product

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latest developments in food authentication A comprehensive overview of authentication techniques and technology Written by an international group of academic and professional peers Provides an excellent complement to more general books on food safety

Federal Student Aid George Scott, 2009-12 The Academic Competitiveness AC and National Science and Mathematics Access to Retain Talent SMART Grants were established by the Deficit Reduction Act of 2005 The grants provide merit based financial aid to certain low income college students eligible for Federal Pell Grants and are administered by the Dept of Education DoE In the first year of implementation participation was lower than expected This report determines 1 factors affecting AC and SMART Grant student participation 2 challenges colleges face in administering the grant programs and 3 the extent to which DoE has assisted states and colleges with implementation Includes recommendations Charts and tables

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