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- (b) Add a suitable scale to the y-axis in Fig. 2.1. Your scale should extend to 2°C above your highest temperature in Table 2.1.

Plot your results from Experiments 1 to 6 on the grid.

Draw two straight lines through your points, one through the first three points, and one through the last three points. Extend your straight lines so that they cross.

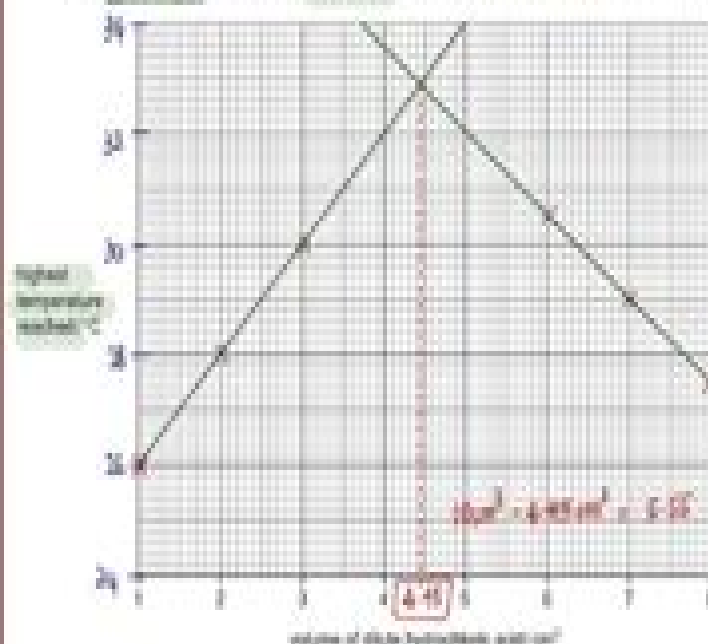


Fig. 2.1

11

- (c) The point on the graph where the two straight lines cross is where all of the aqueous sodium hydroxide reacts with all of the dilute hydrochloric acid to form a neutral solution.

- (i) Use your graph in Fig. 2.1 to deduce the volume of dilute hydrochloric acid and the volume of aqueous sodium hydroxide that react together to produce a neutral solution. Show your working on Fig. 2.1.

volume of dilute hydrochloric acid = 4.4 cm^3 ↓

0620 62 O N 13

JS Bruner



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Unveiling the Magic of Words: A Review of "0620 62 O N 13"

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