



Name: _____ Date: _____

Student Exploration: Nuclear Decay

Background: Alpha particles, beta particles, gamma rays, and neutrons are particles that are emitted from unstable nuclei. They are emitted from the nucleus of an atom and travel through space until they are absorbed by another nucleus.

Prior Knowledge Questions: (Write or tell what you know about each question.)

1. What are some particles, atoms, molecules, and compounds that are made of atoms? (Write or tell what you know about each question.)

Particle	Charge	Mass	Approximate range
Alpha	$+2$	4 amu	10 cm
Beta	-1	0.5 amu	1 m
Gamma	0	0 amu	10 km

1. The **atomic number** of an atom is equal to the number of protons in the nucleus.

Alpha particles have 2 protons and 2 neutrons. What is the atomic number of the parent? (Write or tell what you know about each question.)

2. The **mass number** of an atom is equal to the sum of the number of protons and neutrons in the nucleus. All isotopes of an element have the same atomic number but different mass numbers. What is the mass number of the parent? (Write or tell what you know about each question.)

Alpha Particles

Alpha particles are made of two protons and two neutrons. They are emitted from the nucleus of an atom and travel through space until they are absorbed by another nucleus.

Background: The decay of radioactive isotopes is a process that is used to date objects. The decay of radioactive isotopes is a process that is used to date objects.

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2. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the parent? (Write or tell what you know about each question.)

Background: The decay of radioactive isotopes is a process that is used to date objects. The decay of radioactive isotopes is a process that is used to date objects.



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Richard Bailey



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