

HIMACHAL PRADESH BOARD OF SCHOOL EDUCATION, DHARAMSHALA
Model Question Paper
Class – 11th

Duration – 03:00 Hr

Chemistry(Theory)

M.M.: 60

Instructions:

- i) The question paper consists of four sections- A, B, C and D.*
- ii) Internal choices are given in some questions.*
- iii) Section A contains 12 MCQ 1 to 12 of 1 mark each.*
- iv) Section B contains 9 questions from 13 to 21 of 2 marks each.*
- v) Section C contains 6 questions from 22 to 27 of 3 marks each.*
- vi) Section D contains 3 questions from 28 to 30 of 4 marks each.*
- vii) Make neat and clean diagrams where required.*

Section A (1X12)

Q.1 One mole of oxygen molecule is equal to

- a) 6.023×10^{23} atoms of oxygen b) 6.023×10^{23} atoms of oxygen
c) 6.023×10^{23} molecules of oxygen d) 6.023×10^{23} molecules of oxygen

Q.2 3d orbitals have values

- a) $n=2, l=3$ b) $n=3, l=0$
c) $n=3, l=1$ d) $n=3, l=2$

Q. 3 Which of the following elements has maximum electron gain enthalpy

- a) F b) Cl c) Br d) I

Q.4 Pyramidal geometry associated with

- a) CH_4 b) NH_3 c) H_2O d) CO_2

Q.5 Decreasing order of bond angle in H_2O , NH_3 and CH_4

- a) $\text{CH}_4 > \text{NH}_3 > \text{H}_2\text{O}$ b) $\text{NH}_3 > \text{H}_2\text{O} > \text{CH}_4$
c) $\text{CH}_4 > \text{H}_2\text{O} > \text{NH}_3$ d) $\text{H}_2\text{O} > \text{CH}_4 > \text{NH}_3$

Q.6 According to Boyle's law

- a) At constant T, $V \propto P$ b) At constant T, $V \propto 1/P$
c) At constant P, $V \propto T$ d) At constant T, $P \propto T$

Q.7 For the reaction $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g})$

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RD Boyd



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1. Understanding the eBook Answers For Exercise Of Hsc 11th Chemistry
 - The Rise of Digital Reading Answers For Exercise Of Hsc 11th Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Exercise Of Hsc 11th Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Exercise Of Hsc 11th Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Exercise Of Hsc 11th Chemistry
 - Personalized Recommendations
 - Answers For Exercise Of Hsc 11th Chemistry User Reviews and Ratings
 - Answers For Exercise Of Hsc 11th Chemistry and Bestseller Lists
5. Accessing Answers For Exercise Of Hsc 11th Chemistry Free and Paid eBooks
 - Answers For Exercise Of Hsc 11th Chemistry Public Domain eBooks
 - Answers For Exercise Of Hsc 11th Chemistry eBook Subscription Services
 - Answers For Exercise Of Hsc 11th Chemistry Budget-Friendly Options
6. Navigating Answers For Exercise Of Hsc 11th Chemistry eBook Formats

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 - Answers For Exercise Of Hsc 11th Chemistry Compatibility with Devices
 - Answers For Exercise Of Hsc 11th Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Exercise Of Hsc 11th Chemistry
 - Highlighting and Note-Taking Answers For Exercise Of Hsc 11th Chemistry
 - Interactive Elements Answers For Exercise Of Hsc 11th Chemistry
 8. Staying Engaged with Answers For Exercise Of Hsc 11th Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Exercise Of Hsc 11th Chemistry
 9. Balancing eBooks and Physical Books Answers For Exercise Of Hsc 11th Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Exercise Of Hsc 11th Chemistry
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answers For Exercise Of Hsc 11th Chemistry
 - Setting Reading Goals Answers For Exercise Of Hsc 11th Chemistry
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answers For Exercise Of Hsc 11th Chemistry
 - Fact-Checking eBook Content of Answers For Exercise Of Hsc 11th Chemistry
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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